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3 4	Spatial variation of trace metals within intertidal beds of native mussels (<i>Mytilus edulis</i>) and non-native Pacific oysters (<i>Crassostrea gigas</i>): implications for the food web?
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12	Keywords: Pacific oyster (Crassostrea gigas); Mussel (Mytilus edulis); Non-native species; Invasive species
13	Metals (Cd,Pb,Cu,Zn); North Sea; Trophic transfer; Multiple stressors
14	
15	Abstract
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17	Pollution is of increasing concern within coastal regions and the prevalence of invasive species is also rising. Yet
18	the impact of invasive species on the distribution and potential trophic transfer of metals has rarely been
19	examined. Within European intertidal areas, the non-native Pacific oyster (Crassostrea gigas) is becoming
20	established, forming reefs and displacing beds of the native blue mussel (Mytilus edulis). The main hypothesis
21	tested is that the spatial pattern of metal accumulation within intertidal habitats will change should the abundance
22	and distribution of C.gigas continue to increase. A comparative analysis of trace metal content (cadmium, lead,
23	copper and zinc) in both species was carried out at four shores in south-east England. Metal concentrations in
24	bivalve and sediment samples were determined after acid digestion by inductively coupled plasmaoptical
25	emission spectrometry. Although results showed variation in the quantities of zinc, copper and lead (mg m ⁻²) in
26	the two bivalve species, differences in shell thickness are also likely to influence the feeding behaviour of
27	predators and intake of metals. The availability and potential for trophic transfer of metals within the coastal food
28	web, should Pacific oysters transform intertidal habitats, is discussed.

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33 Introduction

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35 The ecological impact of invasive species is a global issue that has resulted in significant displacement of native 36 fauna and economic impact (Rilov & Crookes, 2009). Increasing temperatures as a consequence of climate change 37 heightens the risk of biodiversity loss from non-native introductions, particularly in temperate and Ppolar 38 Regions (Sorte et al. 2010; de Rivera et al. 2011). Most studies of the impact of non-native species on 39 biodiversity focus on direct ecological effects, such as interspecific competition with native species (see Byers et 40 al. 2009 for review), predator prey interactions (Rilov, 2009), ecosystem engineering (Ruesink et al. 2005; Crooks, 2009) and relative tolerance to environmental stress (Lenz et al. 2011). Amongst many anthropogenic 41 42 impacts that threaten biodiversity of the coastal zone, pollution can also be highly significant (Clark, 2001). The 43 impact of metals from mining operations, domestic discharges and as a component of anti-fouling paints on 44 individual species and species diversity has received widespread attention (Kennish, 1997; Kushel & Timperley 45 1999; Clark, 2001). Tolerance of non-indigenous species to metal--contaminated habitats may have facilitated 46 their spread (Piola & Johnston, 2009; McKenzie et al. 2012). Yet the impact of invasive species on the 47 distribution, and potential trophic transfer, of metals and other pollutants has rarely been examined. Buddo et al. 48 (2012) considered human health implications from harvesting the invasive Indo-Pacific green mussel Perna 49 viridis in the Caribbean and Hübner et al. (2010) investigated the potential release of cadmium from the erosion of 50 accumulated sediments within swards of the invasive cordgrass-grass Spartina anglica. In freshwater systems the 51 invasive macrophyte Eichhornia crassipes (water hyacinth) is known to accumulate considerable amounts of 52 organic and metal contaminants (see Villamagna & Murphy (2010) for review).

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Trace metals are naturally found within the environment, although levels may be elevated within industrial areas, ports and marinas. For example copper is a constituent of antifouling paints and zinc is a prominent constituent of anti-corrosion coatings and urban road run-off due to vehicle tyre wear (Councell et al.2004). Other metals, including cadmium and lead, are used in a variety of industrial processes including colouring, plating and printing; (Guéguen et al. 2011). Some trace metals (e.g. copper and zinc) are considered essential for many invertebrates and have a vital physiological role (Depledge & Rainbow, 1990), yet others (e.g. cadmium and lead) are non60 essential and need to be detoxified or excreted by organisms to avoid toxic effects (Phillips & Rainbow, 1989; 61 Depledge and Rainbow, 1990; Rainbow, 2002). Even essential metals can become toxic to organisms if uptake, 62 from either solution or in food, exceeds the combined rate of excretion and detoxification to reach and then rise 63 above levels that can be tolerated (Bryan, 1971; Rainbow, 2002). The uptake of trace metals by invertebrates may 64 occur directly across permeable membranes from the surrounding seawater, ingestion from water, suspended 65 organic matter and sediments, from food and also from metal-rich particles via pinocytosis (Depledge & Rainbow, 66 1990; Langston et al. 1998). Uptake may be influenced by a number of factors including temperature, salinity, pH 67 and the particular metal species (Depledge & Rainbow, 1990). For the oyster Crassostrea gigas, dissolved copper 68 in ambient water, as opposed to copper in phytoplankton food and sediments, has been shown to be the most 69 important source of copper in tissues (Ettajani et al. 1992).

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71 Bivalves exhibit a range of behaviours and strategies that deal with the potential toxic effects of metals. Trace 72 metals can be excluded from ingestion and be removed in the pseudofaeces of oysters and mussels (Newell & 73 Jordan, 1983; Arifin & Bendell-Young, 1997). The regulation of metal concentrations as a detoxification strategy, 74 where whole body concentrations remain relatively constant and independent of the differences in bioavailability 75 at different sites, requires verification by experiments (Depledge & Rainbow, 1990; Rainbow & Dallinger, 1993). 76 However, partial regulation of zinc has been observed in the mussel Mytilus edulis (Lobel et al. 1982). Oysters, by 77 contrast are generally strong net accumulators of zinc (Rainbow, 1992; Wang & Rainbow, 2008; Giltrap et al. 78 2013). Accumulated metals may be divided in-to that which is metabolically available and a detoxified component 79 that is stored in tissues (Rainbow, 2007). The immobilisation and storage of accumulated and potentially toxic 80 metals is important in both oysters and mussels. However the physico-chemical form of storage can vary across 81 different tissues. For example metals can be stored as insoluble granules within lysosomes in the kidney 82 (M.edulis), and in the digestive gland and basal lamina in both oysters and mussels. Almeida et al. (1998) found 83 evidence of a possible regulatory mechanism in *C.gigas* for lead, whereby excess is directed to the shell through 84 the mantle. Some metals (e.g. Cu and Zn) can be compartmentalised and detoxified in the soluble phase by 85 binding to proteins known as metallothioneines (Phillips & Rainbow, 1989; Rainbow, 2002; Rainbow et al. 2007, 86 2011). These processes can render both Cu and Zn inactive within intracellular metabolic processes (Geffard et al. 87 2004; Rainbow, 2007). The accumulation of metals and other chemicals in molluscs are often measured as part of 88 widespread national and international monitoring programmes such as 'Mussel Watch' (e.g. Tripp, 1992). In 89 addition to the efficiency of detoxification mechanisms, the accumulation of metal in marine bivalves is 90 influenced by a range of other factors. These include the size, age and reproductive condition of the animals 91 (Langston et al. 1998; Bryan et al. 1985), uptake and efflux rates (Wang & Rainbow, 2008) and environmental
92 factors including; salinity, the level of contamination and diet (Roesijadi & Robinson, 1994; Geffard et al. 2004;
93 Lekhi et al. 2008; Wang & Rainbow, 2008).

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95 The trophic transfer of trace metals along food chains is common in aquatic environments (Wang, 2002; Rainbow, 96 2002, Rainbow et al. 2007, 2011; Rainbow & Smith, 2010; Guo et al. 2013). Yet transfer is controlled not only by 97 the concentrations of the metal but also by the food characteristics and physico-chemical form of the metal in the 98 prey, detoxification processes and the digestive and feeding physiology of the predator (Wang, 2002). Therefore 99 significant changes in the uptake of metals from organisms in the wild, and any consequential toxic effects via 100 trophic transfer, will depend on the physiology and tolerance levels of predatory species within the particular 101 ecosystem. Invertebrates from the same site will differ both in their total metal concentrations and partitioning of 102 metals in tissues (Rainbow et al. 2007, 2011; Guo et al. 2013). The trophic availability of metals in prey can be 103 dependent on the digestive and assimilative abilities of the specific predatory species (Rainbow & Smith, 2010).

104 Whelks and other gastropods can be important predators in intertidal benthic environments and the uptake of 105 metals from prey is the most important route of transfer (Blackmore & Wang, 2004; Guo et al. 2013). 106 Investigations of trophic transfer of metals between the rock oyster Saccostrea cucullata and whelk Thais 107 clavigera (Blackmore & Wang, 2004) showed not only that metals were transferred between the trophic levels but 108 also that cadmium, mercury and zinc had high potential for biomagnification in the food chain of rocky shores. In 109 laboratory experiments, over eight weeks there was a higher uptake of zinc, copper and cadmium by the 110 gastropod Nassarius siquijorensis feeding on the oyster Crassostrea angulata compared withte other bivalves 111 (Guo et al. 2013).

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The abundance and spatial extent of organisms that are accumulating metals and their importance to predators will also determine what, if any, influence metals may have on the food web. Studies that describe broad-scale patterns of bioaccumulation and availability in different organisms are required. This may be particularly important for invading species that can often reach high densities.

Introductions of non-native molluscs have been relatively frequent and several species now form important populations in coastal ecosystems (Eno et al. 1997; Miller et al. 2007). Pacific oysters are one of the most 'globalised' marine invertebrates, having been introduced for aquaculture and cultivation into 66 countries outside of-their native range (Ruesink et al. 2005). In several temperate regions, rising sea temperature has resulted in successful spawning and subsequent wild settlement beyond cultivated areas (Reise 1998; Dutertre et al. 2010;

Ruesink et al. 2005, Ruesink 2007). Pacific oysters will settle preferentially on conspecifics and in parts of Europe, intertidal reefs consisting of a dense concretion of oysters have now developed on previously muddy and rocky habitats (Reise, 1998; Lejart & Hily, 2011). In the Wadden Sea, Pacific oyster reefs now occupy large areas of former beds of the native mussel (*Mytilus edulis*) (Diederich, 2005; Nehls & Buttger, 2007) and there are now very few mussel beds without *C.gigas*. In the past decade, the abundance of wild *C.gigas* has increased on the south coast of England and smaller reefs and dense aggregations are occurring locally on a range of substrata including mussel beds (Herbert et al. 2012).

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130 Due to efficient detoxification mechanisms and high uptake rate (Rainbow, 1992; Roesijadi, 1996) cultivated 131 oysters (Crassostrea spp.) are known to accumulate particularly high concentrations of some metal elements such 132 as zinc (Rainbow, 1992; Wang & Rainbow, 2008; Giltrap et al. 2013). It is reasonable therefore to hypothesise 133 that non-native wild C.gigas will also similarly accumulate metals and that the magnitude of accumulation may 134 differ from field measurements in native bivalves. Should C.gigas displace native mussel beds then the spatial 135 distribution and availability of metals across intertidal habitats may change. Intertidal mussels can be important 136 prey for invertebrates including shore crabs (Carcinus maenas) (Dare et al. 1983; Frandsen & Dolmer 2002), dog 137 -whelks (Nucella lapillus) (Crothers, 1985), starfish (Asterias rubens) (Dolmer, 1998) and shorebirds including 138 Oystercatcher (Haematopus ostralegus) (Goss-Custard, 1996; Scheiffarth et al. 2007), Herring gull (Larus 139 argentatus) and Eider duck (Somateria mollissima) (Scheiffarth et al. 2007). In North America, crabs, whelks and 140 sunstars are known predators of C.gigas at aquaculture sites (Quayle, 1964). In laboratory experiments, Dare et al. 141 (1983) observed predation of C.gigas of up to 55-60_mm by large shore crabs (C. maenas), yet stated that most 142 crab predation at aquaculture sites is likely to be on smaller oysters (40-45_mm). Little is known about bird 143 predation on wild Pacific oysters, however Oystercatchers have been observed taking C.gigas at shell lengths 144 between 17-68 mm, though the size range was dependent on season (Markert et al. 2013). Herring gulls are also 145 known to feed on wild Pacific oysters (Scheiffarth et al. 2007; Markert et al. 2013).

The main hypothesis tested therefore is that the spatial pattern of metal accumulation in intertidal habitat will change should wild *C.gigas* increase and displace native fauna. The main objectives are (i) to measure the concentration of metals (Cd, Pb, Cu, Zn) within *M.edulis* and *C.gigas* populations on rocky shores where the two species currently coexist and (ii) to predict changes in their spatial distribution should *M.edulis* be displaced by

- *C.gigas.* Implications of displacement of *M.edulis* by *C.gigas* for the potential trophic transfer of metals in thecoastal food web are discussed.
- 152 Materials and Methods
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- 154 Study area and sampling sites

155 In Kent, south-east England, four shores on the open coast were selected where Pacific oysters have colonised 156 over the past decade (Fig. 1). Epple Bay, Nayland Rock and Foreness Point consist of chalk bedrock with adjacent 157 sand and muddy sediments, whereas Longrock is a mixed sediment shore with flint and shell debris. The mouth of 158 the Thames Estuary that flows through the port of London is about 30 km west of Longrock, yet most of the local 159 coastal land use is suburban residential housing and tourist development. A Pacific oyster farm and hatchery is 160 located 9 km east of Whitstable and the nearest port is at Ramsgate, where a dredge spoil dumping site is located 161 offshore. Salinity is 35 psu and mean sea temperature range is 5–18 °C. Tidal range is 5 m and residual currents 162 run west to east from the Thames estuary along the north Kent coast and north-east from Ramsgate. The upper 163 part of each shore is mostly colonised by fucoid algae, yet mussel beds (M. edulis) dominate large patches of the 164 middle and lower shore. Pacific oysters can be found across the whole intertidal zone, especially on the middle 165 and lower shore amongst the mussel beds. Peak densities of live ovsters recorded at the four sampling sites (Fig.1) 166 during baseline surveys between July 2007 and July 2008 (McKnight, 2012) were Longrock (6 ind. m⁻²), Epple Bay (181 ind. m⁻²), Naylands Rock (44 ind. m⁻²) and Foreness Point (32 ind. m⁻²). At Epple Bay, densities were 167 168 sufficiently high to form reef-like aggregations of contiguous oysters. In many of these aggregations, oysters were 169 vertically orientated and some individuals attain shell lengths >10_cm. The sampling sites to the east of Longrock 170 are within part of the North East Kent European Marine Site (NEKEMS), which has been designated as Special 171 Area of Conservation (EU Habitats Directive) and Special Protection Area (EU Birds Directive).

172

173 Field sampling

^{Samples of Pacific oysters, mussels and sediments were collected in April 2012. Thirty adult} *C.gigas* and *M.edulis*were collected at Mean lLow wWater from each shore, with fifteen individuals taken in two patches at least 100 m

177 apart. Mussel length was between 25-35 mm, which from population size-frequency histograms would indicate 178 they were at least one year of age (Wright & Bailey, 2009). Oysters sampled were from the most frequent size 179 classes (50-70 mm) which from population size--frequency histograms (McKnight & Herbert, unpublished data) 180 are also estimated to have recruited at least one to two years previously. Samples were frozen at -20-°C in labelled, 181 sealed plastic bags. Data on the size, density and biomass of two intertidal mussel beds sampled in 2008 at Shell 182 Ness and Pegwell (Fig 1) were obtained from Wright & Bailey (2009). These estimates of mean biomass and density were based on monthly grab -samples (n =5) of 0.1 m^{-2} . For metal analysis, three surface sediment 183 184 samples (upper 5_cm) of ~250_g were taken at each site within 1_m of the shellfish samples, placed in a labelled 185 bag and frozen. These samples together with a field assessment of the proportion of gravel, sand, silt and clay at 186 each sampling site were used to classify beach sediments according to Folk (1954).

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188 Laboratory analysis

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Defrosted shellfish were opened with a knife to remove the flesh. Oyster and mussel tissues -were removed from shells and dried at 60_-80°C until constant and mean dry weight was calculated. The dried shellfish were weighed and homogenised separately using a mortar and pestle. Sub-samples of invertebrate material (0.200_g +/- 0.001) were weighed into digestion vessels and 1 ml 37% HCl and 4_ml 70% HNO₃ were added. After microwave digestion (Anton Paar Multiwave 3000; 5 min at 750_W, then 25 min at 100_W; Maichin et al. 2000) samples were filtered through Whatman No. 42 filter papers before being made up to a final volume of 50_ml using deionised distilled water.

Sub-samples of sieved sediments (0.300_g +/- 0.001 of particle size < 212 μm) were then weighed into digestion
vessels and 6 ml of *aqua regia* was_added (Millward & Kluckner, 1989). *Aqua regia* is a widely used and
accepted extract for the determination of the pseudo-total metal concentration in sediments and weaker extractants
may fail to dissolve all metals from anthropogenic sources (Peña-Icart et al., 2011).

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After microwave digestion (20 minutes at 200-°C) samples were again filtered through Whatman No. 42 filter papers before being made up to a final volume of 50 ml with deionised distilled water. All equipment for digestion

204	and filtering was acid conditioned in 10 % HNO3. Fisher Primar Plus trace metal grade reagents were used
205	throughout. Metal concentrations in digested samples were determined by inductively coupled plasmaoptical
206	emission spectrometry (ICP-OES) for metal analysis (Varian Vista-Pro in axial configuration). Analytical validity
207	was ensured by the digestion and analysis of certified reference materials (NWRI TH-2 harbour sediment) and
208	process blanks. Recoveries were between 70-80% and precision (%RSD) was between 2.6-3.8 (Table S1).
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210	Statistical analysis
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212	The significance of the differences between the main effects of Species and Site were tested using Two-Way
213	ANOVA in 'R' (R Core Team, 2013). Sediment concentrations were not included in the analysis as the principal
214	aim of the investigation concerns the differences in concentrations between the two shellfish species. One-Way
215	ANOVA was used to test for site-specific differences between each combination of metal and species. In all cases
216	the White correction for heteroscedasticity has been used (White, 1980). Multiple comparisons use the sandwich
217	estimators for the covariance matrix. Multiple comparisons (after allowing for unbalanced samples and
218	heteroscedasticity) were adjusted using Tukey's HSD. This allowed robust pairwise comparisons to be made

219 between mean log concentrations of each metal across sites.

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221 Results

222 Using the Folk Classification (Folk, 1954), both Longrock and Epple Bay have sediments consisting 223 predominantly of muds (Longrock, Gravel-sandy Mud ((g) sM); Epple Bay, Muddy- sand (mS)) whereas Nayland 224 Rock and Foreness Point were classified as *slightly gravelly muddy Sand* ((g) mS). Compared withte some other 225 European regions (Reise 1998; Lejart & Hily, 2011), densities of Pacific oysters on most shores along the Kent 226 coast are relatively low (McKnight, 2012). The mean metal concentrations (Cd, Pb, Cu, Zn) in sediments, mussels 227 and oysters ($\mu g g^{-1}$ dry weight) from each shore are presented in Table 1. Observed concentrations of metals in 228 sediments are well below thresholds in all commonly used sediment quality values, as summarised by Hübner et 229 al. (2009). As shown in the boxplots of Figure S1, the range of values of metal concentrations for individual 230 mussels and oysters can be high, particularly lead and cadmium at Epple Bay and Naylands Rock.

Figure 2 shows the Mean \pm 95% Confidence Intervals of metal concentrations (µg g⁻¹ dry weight) within sediment and shellfish at sampling sites on the north coast of Kent, UK, in April 2012. Cadmium -(Cd), Lead (Pb), Copper (Cu)₂ and Zinc (Zn). Bootstrapped 95% confidence intervals were calculated for all cell means using 1000 random draws with replacement using the procedure in the Hmisc package in R (Harrel, 2014). Significant differences (p 0.05) are apparent for most comparisons between mussels and oysters.

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238 Two-Way ANOVA tables are presented in Table S2. The analysis confirmed the significance of the pattern of 239 differences between means shown by Figure 2. Levene's test showed significant heteroscedasticity in the residuals 240 for all metals. Therefore White-corrected covariance matrices were included in all models in order to adjust for 241 the lack of homogeneity. The reported p-values for the main effects of the two--way ANOVAs were based on 242 Type 3 sum of squares for each metal. Sediment was not included in the test which looked at differences between 243 the two species of shellfish. The analysis indicates highly significant effects of species on the concentration of all 244 metals after holding for sitde effects and site:species interactions. The significance of main effects is questionable 245 if interactions are apparent which change the sign of the response. However, inspection of Figure 2 shows clear 246 differentiation of means for the species effect calculated using non-parametric bootstrapping and consistency in 247 the pattern of difference. The significance of the main effect for species is therefore reliable.

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There was no west_east gradient in the concentration of metals within the sediments sampled along the coast, although mean levels of lead, copper and zinc at the two western sites (Longrock and Epple Bay) were approximately double than those measured at the eastern sites (Nayland Rock and Foreness). There was no relationship between concentrations of metals in sediments and those in the shellfish of any species, although this was not statistically tested due to few sampling sites.

254

255 Cadmium

Levels of cadmium did not significantly differ in sediments at the four sites (Table S1). Yet levels were significantly greater in shellfish, except for at Longrock where the difference in levels between sediments and mussels was not statistically significant. There was no significant difference in the levels in Pacific oysters between sampling sites however concentrations were significantly higher than both sediments and mussels at each site. Mean levels in Pacific oysters (2.2 μ g g⁻¹ dry weight), were over three times that in mussels (0.7 μ g g⁻¹ dry weight) and at each site were always significantly higher than mussels. Levels in Longrock mussels were significantly less than at other sampling sites.

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264 Lead

Levels of lead in sediments at the two western sites were almost double <u>of</u> those measured at the two eastern sites, with mean levels at Epple Bay the highest (10.9 μ g g⁻¹ dry weight). In shellfish, mean levels at each site were always less than in the sediments, although at Longrock and Naylands Rock the difference between mussels and sediments was not significant. In contrast to the other metals examined, levels in oysters were always significantly less than in mussels.

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271 Copper

272 In sediments, levels at the two western sites were more than double those of the two eastern sites.

At all sites, levels in Pacific oysters were significantly higher than both mussels and sediments; mean levels in Pacific oysters were over forty--five times greater than in mussels. Pacific oysters collected from Foreness had highest levels of any site (mean = 516 μ g g⁻¹ dry weight) and were significantly greater than the samples from Epple Bay and Nayland Rock.

277

278 Zinc

Significantly higher concentrations occurred in Pacific oysters compared withto both mussels and sediments. Across the study area, mean zinc concentrations in Pacific oysters were almost thirty times greater than in mussels, with highest mean levels measured at Foreness, $(2600 \ \mu g \ g^{-1} \ dry \ weight)$ in the east of the study area. In mussels, levels were significantly less at Longrock compared withto the other sites and there was no significant difference between levels in mussels and sediments at this location.

Mean flesh dry weights of sampled mussels and Pacific oysters were 0.18 and 0.62 g₁ respectively. Local mussel 285 286 beds are of mean density 2475 ind m^{-2} (Wright & Bailey, 2009) and Pacific oysters reef ~ 200 ind m^{-2} 287 (McKnight, 2012). Estimated mean metal concentrations (mg dry weight m⁻²) within both mussel beds and Pacific 288 oyster reefs are presented in Table 2. Although the mean bivalve density and biomass of a mussel bed is 289 considerably higher than that of a Pacific ovster reef, the average accumulation of copper and zinc ($\mu g^{-1} dr y$ 290 weight) is much greater in *C.gigas* compared with to mussels. Yet lead ($\mu g g^{-1}$ dry weight) is higher in mussels than 291 oysters. Quantities of these metals within intertidal habitats are therefore projected to change should C.gigas reefs 292 form and displace the mussel beds. Zinc concentrations (mg m^{-2}) are predicted to be approximately eight times 293 greater within a Pacific oyster reef compared withto a mussel bed and copper (mg m⁻²) is estimated to be thirteen 294 times higher. Yet quantities of lead (mg m⁻²) could be over twenty times less in a Pacific oyster reef compared 295 withto a mussel bed, although cadmium levels are likely to be unchanged.

296

297 Discussion

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299 Observations of relative metal accumulation in oysters and mussels support data obtained elsewhere, albeit mostly 300 from cultivated shellfish (Table 3). Mussels accumulated higher quantities of lead whereas oysters accumulated 301 greater quantities of copper and zinc. The range of metal concentrations obtained for individual bivalves was high, 302 particularly at Epple Bay and Nayland Rock, however this is typically observed at other locations (Bryan et 303 al.1985).- The study area on the north Kent coast is relatively uncontaminated; although levels of zinc, lead and 304 copper were greater in sediments at the two western sites this is most likely because metals were bound to the 305 finer clays and muds sampled at these locations. Much higher levels of zinc, copper and cadmium have been 306 accumulated in Pacific oysters at more contaminated locations (Table 3; Amiard et al. 2008). Although there have 307 been other recent determinations of metals in mussels from this region of South-Eeast England (BODC, 2014), 308 there are no comparative values for wild Pacific oysters available. The levels of each of the four metals found in 309 mussels are comparable with data obtained from shellfish cultivation regions in the UK over the past 20 years 310 (Table 3). Values of lead and cadmium in Pacific oysters are also similar to other determinations, though levels of 311 copper and zinc are higher than previous mean values for England and Wales and the loughs of Northern Ireland

312 (Table 3) (although mean values of zinc in Carlingford Lough obtained between 2010–2012 was 1702 μg g⁻¹ dry
313 wt. (BODC, 2014).

314

315 As the purpose of the investigation was to determine the magnitude of potential metal transfer should the bivalves 316 be consumed by predators within the ecosystem, shellfish were not purged of faeces prior to analysis. Values 317 obtained in this study may therefore be slightly higher than other analyses where bivalves have been purged or 318 undergone depuration. However from estimates of annual biodeposition rates in C.gigas (Nishjkawa-Kjnomura, 319 1978) quantities of Cu and Zn present in the gut is likely to be less than 1% of flesh estimates. Standardisation of 320 metal concentrations to allow for variation in reproductive condition between the two species was not carried out. 321 For mussels, spawning is known to occur during April-May in this region (Wright & Bailey, 2009) and Pacific 322 oysters in summer, assuming sustained temperatures for gametogenesis (Miossec et al. 2009). Bryan et al. (1985) 323 reviewed the influence of sampling period for both species and recommended metals analysis of mussels in late 324 winter to avoid variations in concentration due to food supply and gonad development. In C.gigas, lower metal 325 concentrations in winter and summer are attributable to reduced dry mass and dilution due to gonad development, 326 respectively. In field populations of C.gigas in the Bay of Bourgneuf (France) concentrations of Cd, Cu Zn, Cd 327 and Pb decreased in the late spring and early summer concomitantly with the increase of soft tissue weight and 328 sexual maturity. Inter-annual variations in concentrations of the metals were also associated with weight changes 329 (Amiard & Berthet, 1996). Therefore, taking account of possible variation in the two species in respect of 330 differing reproductive periods our estimates of metal concentrations are likely to be below annual maximum levels 331 for mussels, whereas estimates in oysters are probably closer to the annual maximum. In the Bay of Bourgneuf, 332 Pb and Cd concentrations in oysters were not correlated with the age or weight; however, Cu and Zn 333 concentrations generally did increase with age (Amiard & Berthet, 1996). Concentrations of Cd and Zn measured 334 in sediments and in *M.edulis* (four size classes, 35 to >50 mm) from two estuaries in sSouth-east England 335 decreased between December and August in both the years studied (Wright & Mason, 1999). However, overall 336 differences in mean concentrations of Cd and Zn were much greater than variation attributed to shellfish age and 337 sampling season.

In this study, the considerably high species_-specific differences in the concentration of these particular metals are
likely to be significantly greater than <u>the</u> variations associated with season and size of shellfish and are consistent
with the overall pattern of accumulation of these metals in the two species (Table 3).

341

342 Troost et al. (2009) argue that food intake of wild Pacific oysters may be greater than mussels due to the oyster's 343 high filtration rate and the larger roughness and height of oyster beds, which create more near-bed turbulence and 344 facilitates efficient ingestion. The quantities of zooplankton ingested may also be determined by near-bed 345 roughness differences (Troost et al. 2009). Although untested, turbulence may be different in wild reefs than 346 where *C.gigas* is cultivated in bags on trestles, which may cause differences in accumulated metals due to varied 347 efficiency and rate of ingestion. Variation in the quantities of accumulated metals within the two species is likely 348 to be the result of different ingestion efficiencies, food type and the ability to regulate different metals. Compared 349 withto oysters, zinc appears to be regulated more efficiently in mussels (Lobel et al. 1982). The ability of *M.edulis* 350 to limit the accumulation of copper is dependent on which tissues and organs are examined, the exposure period 351 and the levels of contamination (Amiard-Triquet et al. 1986). In this study, levels of copper were high in both 352 species and particularly oysters. However the metal regulatory abilities of each species need to be verified.

353

354 The trophic transfer of trace metals along food chains is common in aquatic environments (Rainbow, 2002, 355 Rainbow et al. 2007, 2011; Wang, 2002; Guo et al. 2013). Should C.gigas increase and replace beds of native 356 *M.edulis*, what effect might this have on the trophic transfer of metals in coastal habitats, and what, if any, are the 357 ecological implications? This will depend on whether potential predators can access and assimilate metals within 358 the oysters, the physico-chemical form of the metal in the oysters (Rainbow et al. 2002, 2011), detoxification 359 processes and the digestive and feeding physiology of the predator (Wang, 2002). Although experiments have shown that C.gigas of up to 55-60_mm shell length can be consumed by shore crabs (Carcinus maenas) (Dare et 360 361 al. 1983), significant predation by invertebrates of adult C.gigas in the North Sea has not been observed (Troost, 362 2010), presumably due to the thickness of shells. Oystercatchers (Haematopus ostralegus) may prise_off and 363 consume smaller Pacific ovsters (Markert et al. 2013), however their abundance on former mussel beds in the 364 Wadden Sea has significantly declined since colonisation by C.gigas (Scheiffarth et al. 2007).

Little predation was observed during experiments on *C.gigas* interactions on rocky shores (Ruesink, 2007). Therefore, although Pacific oysters can accumulate larger quantities of copper and zinc than mussels, because adult oysters are relatively inaccessible to predators, the displacement of mussel beds and other intertidal habitats is unlikely to increase the trophic transfer of these metals to wild predators within the ecosystem. On the contrary, the transfer of lead to predators, such as the gastropod *Nucella lapillus* (Crothers, 1985), may actually reduce if mussels cannot be exploited and predators switch to alternative food sources.

372

373 Spatial and temporal variation in abundance and the extent to which mussels and oysters are significant prev is 374 important when considering trace metal availability to the coastal food web. Inter-annual recruitment of mussels is 375 known to vary considerably (Dare et al. 2004) and therefore the size and density of intertidal beds will also 376 change. The size -frequency of the mussel population varies throughout the year and it is known that 377 concentrations of metals can fall with increasing weight (Phillips, 1980). Although dense aggregations and small 378 reefs of Pacific oysters are now occurring in southern England, the extent to which *C.gigas* will persist, cause 379 extensive reef formation and thus change the intertidal habitat will depend on the magnitude and frequency of 380 recruitment and mortality. There is strong evidence to suggest that gametogenesis, spawning, larval development 381 and survival are determined by the attainment of threshold temperatures (Miossec et al. 2009) which have 382 increased in frequency within the past two decades (Dutertre et al. 2010). Recently, native species of rocky shores 383 at the northern edge of their biogeographical range have spread in response to warming (Hawkins et al. 2009), so 384 it is likely that *C.gigas* will also spread from established strongholds (Pinnegar et al. 2012). Unlike mussels, 385 C.gigas can potentially occupy all levels of the shore and have been found at Mean High Water on sea walls and 386 across other habitats including intertidal beds of the polychaetes Sabellaria spinulosa and Lanice conchilega 387 (McKnight 2012). The potential change in metal distribution across the shore could therefore be much greater 388 should the abundance of Pacific oysters increase as predicted. Although observed in the Wadden Sea (Smaal et al. 389 2005), there is little evidence of widespread settlement of C.gigas below the intertidal zone in the UK and 390 therefore subtidal beds of *M.edulis* are likely to be unaffected. Subtidal *M.edulis* may resupply affected intertidal 391 areas and enable survival of mussels, at least in patches. In the Dutch Wadden Sea, some recovery of mussel beds 392 has been observed (Nehls & Büttger 2007) and mussels have been found to colonise the interspaces between 393 oyster shells, so some intertidal co-existence appears likely even if C.gigas becomes dominant (Diederich, 2005; 394 Eschweiler & Christensen 2011).

396 Although untested, it is possible that elevated levels of particular metals will occur in Pacific oyster biodeposits, 397 including faeces and pseudofaeces within the sediment surrounding the oysters, as shown under experimental 398 conditions (Nishjkawa-Kjnomura, 1977). If this is evident in the wild, constituent metals could be passed along 399 food chains as the invertebrate diversity amongst the shell interspaces and in sediments below the reefs of C.gigas 400 can be higher than the surrounding sediments (Lejart & Hily, 2011). Potentially at least, prey items within 401 biodeposits amongst the oyster interspaces might be exploited by foraging avian predators (Escapa et al. 2004). 402 The possibility for changes in trophic transfer of metals may therefore extend beyond the individual Pacific 403 oysters themselves and warrants further investigation. Bryan & Langston (1992) specifically highlighted the need 404 for more studies on the impacts of elevated levels of inorganic and organometals on coastal birds.

405

Compared withte other regions of the world, the north coast of Kent is relatively uncontaminated and wild harvesting of Pacific oysters for human consumption is not currently widespread, though is increasing in areas where they are forming dense aggregations. Maximum permitted levels of metals within food for human consumption can vary between countries (Amiard et al. 2008; Guéguen et al. 2011), however maximum concentrations of the bioavailable component of both essential and non-essential metals in *C.gigas* have been recorded above food safety limits in contaminated regions (Bragigand et al. 2004; Amiard et al. 2008; Osuna-Martínez et al. 2011; He & Wang, 2013).

413

- 414 Conclusion
- 415

The main hypothesis tested, that the spatial pattern of metal accumulation within intertidal habitats will change should the abundance and distribution of wild *C.gigas* continue to increase and displace *M.edulis*₂ is supported by our analysis. Quantities of zinc and copper and lead (mg m⁻²) potentially available for consumption by predators are likely to change significantly. Yet, notwithstanding differences in <u>the</u> bioaccumulation of metals in the two bivalves and their potential availability, any broad-scale changes in trophic transfer of metals will also be dependent on whether predators are able to co-exist with and exploit the new invader for food. If the invading

422	species and native species are able to co-exist then any change in trophic transfer is likely to be affected by the
423	selective feeding behaviour of the predators and the relative changes in density of interacting species.
424	There remains significant uncertainty about the ecological effects of metals on aquatic assemblages (Langston et
425	al, 1998; Mayer-Pinto et al. 2010). The potential for significant spatial variation in trophic transfer of metals as a
426	result of invasions of non-native species certainly exists. Whether this has any significant impact ecologically or
427	on human health is also likely to be dependent on local levels of contamination.
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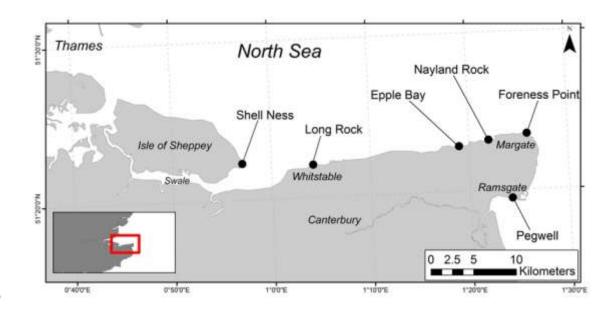
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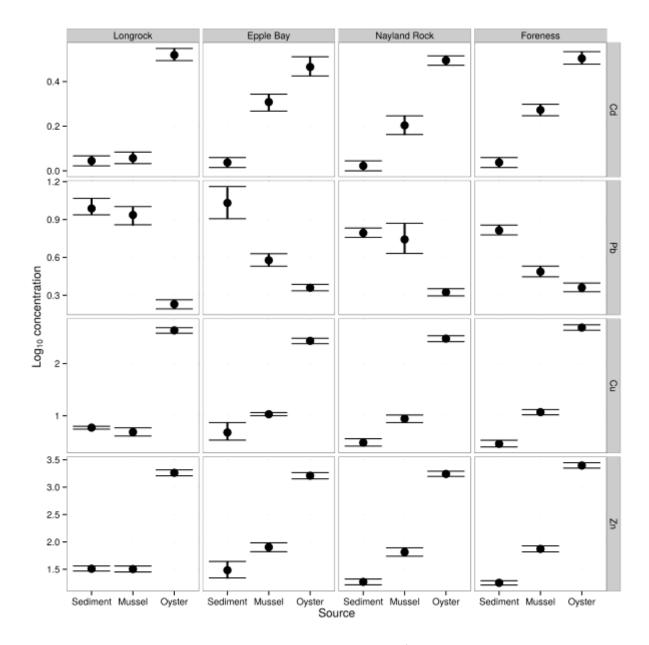


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Fig 1. The Thanet coast located in Kent (south-east England), showing mussel and oyster sampling sites at Long
Rock, Epple Bay, Nayland Rock and Foreness Point. Estimates of mussel density and biomass were obtained from
Shell Ness and Pegwell (Wright & Bailey, 2009). The main port is at Ramsgate.

694



700Fig 2. Mean \pm 95% Confidence Intervals of metal concentrations (µg g⁻¹ dry weight) within sediment and shellfish701at sampling sites along the north coast of Kent, UK, in April 2012. -Cadmium -(Cd); Copper (Cu); Lead (Pb) and702Zinc (Zn). Significant differences are apparent for most comparisons between the shellfish species.

	Sediments				Sediments M.edulis						C.gigas				
	Longrock	Epple	Nayland s	Foreness	Mean	Longrock	Epple	Nayland s	Foreness	Mean	Longrock	Epple Bay	Nayland s	Foreness	Mean
		Bay	Rock		(SD)		Bay	Rock		(SD)			Rock		(SD)
Cd	0.1	0.1	0.1	0.1	0.09	0.2	1.1	0.7	0.9	0.7	2.4	2.1	2.1	2.2	2.19
	(0.08)	(0.09)	(0.08)	(0.09)	(0.09)	(0.22)	(0.49)	(0.44)	(0.33)	(0.37)	(0.62)	(1.07)	(0.63)	(0.61)	(0.73)
Pb	9.1	10.9	5.3	5.6	7.7	8.3	3.0	6.9	2.2	5.1	0.7	1.3	1.1	1.4	1.14
	(3.46)	(5.51)	(0.89)	(0.98)	(2.71)	(2.91)	(1.45)	(7.86)	(1.06)	(3.32)	(0.41)	(0.38)	(0.41)	(0.58)	(0.44)
Cu	5.00	4.80	2.2	2.0	3.48	4.50	9.9	8.7	11.30	8.59	453.80	285.70	309.30	515.70	391.36
	(0.61)	(3.97)	(0.92)	(0.76)	(1.56)	(2.77)	(2.23)	(4.10)	(3.47)	(3.14)	(158.15)	(88.16)	(138.79)	(187.83)	(143.23)
Zn	31.80	34.5	17.9	17.1	25.32	35.0	91.30	71.9	77.9	69.06	1912.80	1712.1	1663.7	2600.0	1972.17
	(6.61)	(20.66)	(3.87)	(2.72)	(8.46)	(21.71)	(52.29)	(32.10)	(26.30)	(33.10)	(544.07)	(533.20)	(535.53)	(855.35)	(617.04)

Table 1: Mean metal concentrations ($\mu g g^{-1}$ dry weight) and $\pm SD$ (italics) in each source from each shore

716	Table 2. A comparison of estimated mean metal concentrations (mg dry weight m ⁻²) obtained in this study within
717	an intertidal mussel bed (mussel density 2475 ind. m ⁻² , Wright & Bailey, 2009) and Pacific oysters reef (oyster
718	density ~ 200 ind. m ⁻² (McKnight, 2012). The mean metal concentrations (Cd, Pb, Cu, Zn) in oysters and mussels
719	($\mu g g^{-1}$ dry weight) from each shore are presented in Table 1. See text for further details.

	Cadmium	Lead	Copper	Zinc
	(mg dry weight m ⁻²)			
Mussels (M.edulis)	0.3	2.3	3.8	30.8
Wwild Pacific oysters (<i>C.gigas</i>)	0.3	0.1	48.5	244.5

Table 3. Mean metal concentration (µg g⁻¹dry weight) in *M.edulis* and *C.gigas* from UK and France. Samples of *C.gigas* used in this study were from wild populations.

723	Contaminated samples from Ronce les Bains and the	Gironde Estuary were obtained in	1990s (Bragigand et al. 2004, Table 1).

	Mussels (Myt	ilus edulis)			Pacific oyster	s (Crassost			
Location		Lead			Cadmium	Lead	Copper	Zinc	References
	$\mu g g^{-1}$								
North Sea,	0.7	5.1	8.6	69	2.2	1.1	391	1972	This study
Kent									
Irish Sea	1.3	1.9	6.7	91	1.2	0.9	91	1196	BODC, 2014.
(Mean 2008-									
2012). ^a									
England &	1.3	5.5	9.0	110	1.05	1.0	208	1000	Jones & Franklin,
Wales -(Mean									2000.
1995 <u>–</u> 1996). ^a									
Atlantic,	-	-	-	-	2.0	-	100	2237	Sources in Table 1 of
Arcachon Bay,									Bragigand et al.
France									2004.
Atlantic, Pen	-	-	-	-	0.7	-	73	1418	Sources in Table 1 of
Be, France									Bragigand et al. 2004.
Atlantic,	-	-	-	-	2.3	-	210	2209	Sources in Table 1of
Bourgneuf									Bragigand et al.
Bay, France									2004.
Atlantic, Ronce	-	-	-	-	6.7	-	220	2782	Sources in Table 1 of
les Bain, France									Bragigand et al. 2004.
Atlantic,	-	-	-	-	75	-	1041	4964	Sources in Table 1 of
Gironde									Bragigand et al.
Estuary, France									2004.

^a wet weight data multiplied by 5 to convert to dry weight (OSPAR, 2009; Amiard et al. 2008)