Thermodynamic performance of a novel shell-and-tube heat exchanger incorporating paraffin as thermal storage solution for domestic and commercial applications

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# 1 Abstract

2 This article is focused to evaluate thermal performance of commercial grade paraffin in a novel shell-and-tube heat exchanger with multi-tube passes and longitudinal fins as latent 3 heat storage (LHS) system. Thermal performance assessments of latent heat storage 4 5 system are conducted with respects to charging/discharging power, accumulative thermal 6 energy storage/retrieval, thermal efficiencies and effectiveness. heat transfer 7 characterisation and nature of melt front propagation. The average charging and discharging 8 powers are significantly enhanced by 75.53% and 27.04% with an increase in temperature gradient between paraffin and inlet water from 52 °C – 62 °C and 15 °C – 5 °C, respectively. 9 Likewise, the maximum charging and discharging powers are augmented from 2.15 kW -10 2.63 kW and 5.18 – 10.37 kW with an increase in flow rate from 1.5 – 3 l/min, respectively. 11 Furthermore, the average effectiveness, Nu-Ra and heat transfer coefficient are significantly 12 13 improved with an increase in temperature gradient and moderately reduced with upgrading volume flow rate. The range of Rayleigh numbers for charging cycles have indicated 14 turbulent nature of melt front movement and supportive behaviour of longitudinal fins 15 orientations towards natural convection. Empirical correlations for average effectiveness and 16 17 Nu-Ra are developed from experimental results to facilitate design estimation for employment of proposed LHS systems in domestic and commercial applications. These 18 empirical correlations, to evaluate feasibility and employability of LHS systems in practical 19 applications such as domestic hot water supply and space heating to maintain control room 20 21 temperature, have been successfully implemented in real operating conditions.

22

# 23 Keywords

- 24 Thermal energy storage (TES); Latent heat storage (LHS); Phase change material (PCM);
- 25 Shell-and-tube heat exchange; Natural convection; effectiveness-NTU

26

# Nomenclature

A <sub>ext</sub>	External heat transfer surface area $(m^2)$	$t_c$	Complete charging time $(s)$
С	Constant value	$t_d$	Complete discharging time $(s)$
$C_p$	Specific heat capacity of water $(J/(kg.K))$	$T_f$	Final temperature of charging cycle (°C)
$C_{p,HX}$	Specific heat capacity of heat exchanger $(J/(kg.K))$	$T_i$	Initial temperature of charging cycle (°C)
$C_{p,pcm}$	Specific heat capacity of PCM $(J/(kg.K))$	T <sub>in</sub>	Inlet temperature of water (°C)
$D_{T,in}$	Inner diameter of tube $(m)$	$T_m$	Phase change temperature of paraffin (°C)
g	Gravitational acceleration $(m/s^2)$	$\overline{T_m}$	Average transient temperature of paraffin (°C)
$Gr_L$	Grashof number	$T_{out}$	Outlet temperature of water (°C)
L	Latent heat capacity of PCM $(J/kg)$	$T_s$	Surface temperature of the tube (°C)
$L_c$	Characteristic height (m)	U	Overall heat transfer coefficient $(W/(m^2.K))$
'n	Mass flow rate of water $(kg/s)$	Greek	a letters
$m_{HX}$	Mass of heat exchanger material $(kg)$	α	Thermal diffusivity of paraffin $(m^2/s)$
$m_{pcm}$	Mass of PCM $(kg)$	β	Coefficient of thermal expansion of PCM $(1/K)$
Nu	Nusselt number	ε	Instantaneous effectiveness
$\overline{P_c}$	Average charging power (W)	$\overline{\varepsilon_c}$	Average effectiveness of charging cycle
$\overline{P_d}$	Average discharging power (W)	$\overline{\varepsilon_d}$	Average effectiveness of discharging cycle
Pr	Prandtl number	η <i>c</i>	Charging efficiency
$Q_c$	Thermal energy charged (J)	$\eta_d$	Discharging efficiency
$Q_{c,max}$	Maximum thermal energy charged $(J)$	v	Kinematic viscosity of paraffin $(m^2/s)$
$Q_d$	Thermal energy discharged (J)	$\mu_w$	Dynamic viscosity of water $(Pa. s)$
$Q_{d,max}$	Maximum thermal energy discharged (J)		
$Q_{HX}$	Thermal energy stored by heat exchanger (J)	Acrony	rms
$Q_{PCM}$	Thermal energy stored by PCM $(J)$	LHS	Latent heat storage
Ra	Rayleigh number	NTU	Number of transfer units
Re	Reynolds number	PCM	Phase change material
Ste	Stefan number	TES	Thermal energy storage
$\Delta t$	Time steps to record data (s)		

## 27 1. Introduction

28 The global economic progression and social developments are coupled with higher primary energy supply demands. In the past four decades, the dependency on fossil fuels to meet 29 global energy demands have remained moderately unaffected with an insignificant reduction 30 31 from 86% to 78.4% [1]. However, the excessive reliance on fossil fuels have resulted in 32 depletion of natural resources and emission of hazardous gases to environment. In order to mitigate global warming and climate change challenges, the development and utilisation of 33 34 new technologies for renewable energy sources are extremely crucial [2, 3]. Latent heat 35 storage (LHS) systems coupled with solar energy sources or heat recovery systems are 36 appraised as decisive technology to offset mismatch between thermal energy supply and demand. Energy systems yield comparatively better thermal efficiency and sustainability with 37 the integration of LHS systems. Moreover, LHS systems, with their higher thermal storage 38 39 capacity at almost isothermal condition, wide-ranging availability of phase change materials 40 (PCM), negligible environmental hazards and low vapour pressure encourage practical utilisation in domestic and industrial applications [4-7]. 41

However, the low thermal conductivity of PCMs (~ 0.2 W/m.K) significantly impedes the 42 thermal efficiency and the large-scale practical applications of LHS systems. Hence, thermal 43 44 performance enhancement techniques are developed and implemented by researchers to 45 overcome this deficiency. For instance, geometrical orientation of the container, inclusion of 46 extended surfaces with heat exchangers, addition of thermal conductive additives and encapsulation techniques are extensively examined [8-10]. This article will discuss literature 47 concerning shell-and-tube heat exchangers with extended surfaces due to their relatively 48 49 higher heat transfer potential, minimal design complexity and easier integration to practical 50 applications [11, 12].

Rathod and Banerjee [13] conducted both charging and discharging experimental analyses 51 52 of stearic acid in a vertical shell-and-tube heat exchanger with three longitudinal fins. It was reported that heat transfer performance at varied operating conditions were improved with 53 inclusion of longitudinal fins. Hence, the charging and discharging rate was significantly 54 enhanced by 24.52% and 43.6%, respectively. Li and Wu [14] informed that with inclusion of 55 six longitudinal fins in a horizontal shell-and-tube heat exchanger, the average heat flux was 56 57 significantly enhanced and consequently, the charging and discharging rate was enhanced by 14% as compared to the no-fins configuration. Ye [15] conducted numerical simulation of 58 59 paraffin in horizontal double tube heat exchanger with varying longitudinal fins from 0 – 10. It was reported that as compared to no-fins orientation, the melting time was significantly 60 reduced by 40% - 86% with an increase in longitudinal fins from 2 - 10, respectively. In 61 another study, Ye [16] numerically examined the optimal aspect ratio for melting process of 62 paraffin in vertically heated rectangular container. It was concluded that melting process can 63 64 be augmented with an increase in aspect ratio from 0.1 - 1 and the optimal melting 65 performance can be achieved for aspect ratio  $\geq$  1. Similarly, Rabienataj Darzi et al. [17] numerically investigated the thermal behaviour of n-eicosane in a horizontal shell-and-tube 66 heat exchanger with varying number of longitudinal fins. It was noticed that as compared to 67 68 the no-fins configuration, an increase in longitudinal fins from 4 - 20 significantly enhanced the charging rate by 39% – 82% and discharging rate by 28% – 85%, respectively. Thermal 69 70 performance enhancements with inclusion of longitudinal fins were more effective in 71 discharging cycles due to the fact that higher numbers of extended surfaces obstruct effective natural convection during charging cycles. Moreover, Lohrasbi et al. [18] conducted 72

73 numerical evaluation of thermal performance of PCM in a vertical shell-and-tube heat 74 exchanger with no-fin, longitudinal fin and circular fin orientations. Longitudinal and circular fins augmented the thermal penetration depth and consequently, the phase transition rate 75 was enhanced by 3.26 and 3.55 times, respectively. Caron-Soupart et al. [19] performed 76 77 experimental analyses on paraffin in three configurations of a shell-and-tube heat exchanger with no-fins, longitudinal fins and circular fins. It was reported that longitudinal and circular 78 fins had improved thermal power by a factor of 10. Moreover, the role of natural convection 79 80 on temperature distribution during charging process was significant. Kabbara et al. [20] experimentally examined the charging and discharging behaviour of dodecanoic acid in a 81 vertical shell-and-tube heat exchanger with rectangular fins and four tube passes. It was 82 83 recorded that the average power was merely increased from 0.24 - 0.32 kW with an increase in volume flow rate from 0.7 - 2.5 l/min. It can be noticed from the literature review 84 85 that a single tube pass of a shell-and-tube heat exchanger with longitudinal fins are 86 extensively investigated both numerically and experimentally. However, the literature lacks thermal evaluation of shell-and-tube heat exchangers with multi-tube passes and longitudinal 87 88 fins with relatively higher thermal storage capacity to support integration into practical 89 applications. Therefore, a novel geometrical configuration of a shell-and-tube heat exchanger with multi-tube passes and longitudinal fins was previously designed and 90 optimised in [21], developed and experimentally examined for charging and discharging 91 92 cycles in [22, 23]. The experimental evaluations were focused on vertical and radial temperature distributions, and melting and solidification rates of paraffin in the proposed 93 design solution. 94

95 In past decade, the inclusion of thermal conductive additives are extensively investigated for 96 their potential to improve thermal performance of LHS systems [24]. Yang et al. [25] 97 conducted numerical investigations on metal foam enhanced paraffin in vertical shell-andtube heat exchanger. It was reported that with inclusion of metal foam (copper) in paraffin 98 99 with optimal porosity of 0.94, the melting rate was significantly augmented by 88.54% as 100 compared to no metal foam enhanced paraffin. Yang et al. [26, 27] furthered the numerical analyses by probing the impacts of natural convection and angles of inclination of container 101 on charging rates of metal foam based paraffin composite. It was reported that natural 102 103 convection based numerical simulations had presented significantly higher phase transition 104 rates (45.5%) as compared to conduction dominant simulations. Moreover, due to excellent 105 temperature distribution in metal foam enhanced paraffin, the angles of inclination had demonstrated minimal enhancement of 4.35% between cases with angles 0° and 90°. In 106 other words, the natural convection has minimal influence on phase transition process in 107 metal foam enhanced LHS systems and the uniform temperature distributions were owing to 108 conduction dominance. Mahdi and Nsofor [28] reported similar observations in their 109 numerical examination of metal foam enhanced PCM in horizontal shell-and-tube heat 110 exchanger. Meng and Zhan [29] conducted experimental charging and discharging cycles on 111 112 copper foam enhanced paraffin in vertical shell-and-tube heat exchanger. It was reported 113 that the charging power was improved from 0.13 kW to 0.22 kW with an increase in inlet temperature from 75 °C to 85 °C. Likewise, the discharging power was enhanced from 0.17 114 kW to 0.27 kW with reducing the inlet temperature from 20 °C to 10 °C, respectively. 115 Furthermore, the inclusion of metal oxides (Al<sub>2</sub>O<sub>3</sub>), metal nitrides (AlN) and carbon allotropes 116 (GNP) based nano-additives to paraffin in horizontal shell-and-tube heat exchanger was 117 experimentally and numerically investigated in [30]. It was reported that as compared to pure 118 paraffin, the charging rates for Al<sub>2</sub>O<sub>3</sub>, AIN and GNP based nano-PCMs were significantly 119

enhanced by 28.01%, 36.47% and 44.57%, and the discharging rates by 14.63%, 34.95%
and 41.46%, respectively. However, it was reported in [31-34] that as compared to thermal
conductive additives enhanced LHS systems, the phase transition rates and thermal storage
capacities of LHS systems with optimised extended surfaces were relatively higher.

124 Effectiveness-NTU is considered as a decisive technique for characterising and evaluating heat transfer performance for specific designs of LHS systems [35]. This technique is 125 implemented by researchers to assess and optimise thermal performance of proposed 126 127 design solutions of LHS systems. Tay et al. [36] conducted experimental and numerical analyses on enhancement in average effectiveness and phase change rate of PCM in LHS 128 system with no-fins, pins and circular fins orientations. It was reported that amongst 16 tubes 129 orientations with similar volumes, the tube with circular fins presented higher average 130 effectiveness and rapid phase change rate as compared to no-fins and pins orientations. It 131 was due to relatively higher effective surface area of circular fins which enabled the 132 133 enhancement in average effectiveness and phase change rates by 20 - 40 % and 25 % as compared to no-fins orientation, respectively. Moreover, the empirical correlations for 134 effectiveness as a function of effective surface area and operating mass flow rate of HTF are 135 136 developed in [37-40]. Despite being significantly influential parameters, the literature lacks consideration of operating inlet temperature and phase change temperature of PCM while 137 developing empirical correlations for average effectiveness. Furthermore, the previous 138 literature do not consider in-depth evaluation of design configurations impacts on natural 139 140 convection and nature of melt front propagation.

This article is the continuation of experimental thermal performance evaluation of the novel 141 LHS system in terms of comprehensive transient and average charging/discharging power, 142 accumulative thermal energy storage/retrieval, charging/discharging thermal efficiencies, 143 average effectiveness assessment and natural convection characterisation at varied 144 operating conditions. Thermal performance evaluations of proposed novel LHS system are 145 based on experimental data acquired from twenty-eight charging and discharging cycles, 146 147 which is symbolic and unprecedented contribution to literature. The above-mentioned 148 thermal performance evaluations have not been considered or published in our earlier articles [21-23]. Moreover, this article is focused on developing empirical correlations for 149 charging and discharging effectiveness, which will empower to evaluate the design 150 characterisation and employability of this novel LHS system in practical applications. The 151 empirical correlations are derived which considers parameters for inlet temperature and flow 152 rate of heat transfer fluid (water), phase change temperature of paraffin (or other PCM) and 153 effective external surface area of the heat exchanger. The developed empirical correlations 154 155 are unique in terms of empowering implementation for a range of phase change materials and varied operating conditions of heat transfer fluid, which are not reported in previous 156 literature. Moreover, the average effectiveness of the proposed novel LHS system is 157 compared with other published designs. Similarly, the impact of natural convection on heat 158 transfer and melt front propagation of paraffin in novel design configuration are characterised 159 by evaluating Nusselt number (Nu) and Rayleigh number (Ra). An empirical correlation is 160 developed for Nu-Ra and compared with established results in literature. The practicality and 161 viability of proposed LHS system integration to domestic and commercial applications are 162 evaluated by conducting two case studies. 163

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# 165 2. Experimental setup and procedure

166 2.1. Experimental setup

167 The experimental test rig is demonstrated in a schematic diagram, as shown in **Fig. 1**. The 168 test rig consists of a flat plate solar collector setup with solar simulators, latent heat storage 169 (LHS) system, connection to the municipal water supply, centrifugal pump, flow control 170 valves and data acquisition system with computer.



171

Fig. 1 Schematic representation of experimental test rig for charging/discharging cycles of LHSsystem coupled with flat plate solar collector.

To conduct charging cycles, the water supply is directed to pass through serpentine copper 174 tubing in the solar collector, where steady radiant thermal energy from solar simulators is 175 absorbed by the water. The technical specifications and operation details of the solar 176 177 simulators based solar collector system is discussed comprehensively in [22, 41], as presented in Table 1. The high temperature water at the solar collector outlet is directed to 178 179 pass through the LHS tank, where heat transfer occurs between high temperature water in 180 the tubes and paraffin in the shell container. Thermal energy is transmitted to the paraffin in 181 the shell container and therefore, the low temperature water at the LHS outlet is pumped back to the solar collector to repeat the charging process. Technical design specifications of 182 shell-and-tube heat exchanger with multi-tube passes and longitudinal fins are previously 183 discussed and detailed in [21, 22], as shown in Table 2. Likewise, the physical model of the 184 LHS unit is illustrated in Fig. 2 and the thermo-physical properties of commercial grade 185 paraffin (RT44HC) are presented in Table 3. Technical limitations of lab-scale solar collector 186 system are considered while selecting the appropriate range of phase transition temperature 187 188 of paraffin. Hence, in case of larger solar collector, the paraffin with higher phase transition temperature can be utilised in this novel designed LHS system. 189

190 A Grundfos centrifugal pump (UPS 15-60) is employed to circulate water in a closed loop between the solar collector and the LHS unit. Moreover, four 3-way valves are installed to 191 conduct charging/discharging cycles by adjusting flow direction and operating volume flow 192 rate. FT2 turbine flow meters are utilised to record volume flow rates of the water. 193 Furthermore, to record thermal response of the paraffin in the shell container to inlet 194 temperature of the water, fifteen K-type thermocouples are installed at five zones (A, B, C, D 195 and E) and at three vertical positions (top, central and bottom) at each zone, as illustrated in 196 197 Fig. 2. The vertical positions of thermocouples are at a vertical distance of 115 mm to each 198 other. Also, two K-type thermocouples are coupled to the copper tube at the inlet and outlet of the LHS unit to register temperature response of the water. The accuracy of the turbine 199 flow meters and thermocouples are 1.5% and ±1.5 °C, respectively. An Agilent data 200 acquisition (34972A) unit is connected to the sensors to aquire temperature and volume flow 201 202 rates data. Agilent software is operated to log temperature and volume flow rates data at 203 time steps of 10 s.

Table 1		441
Components	Parameters	, 41] 
Componenta	Length (m)	24
Gross Dimensions	Width (m)	2.4
Cross Dimensions	Thickness (m)	0.254
	Material	Transparent glass
	Thickness (m)	0.002
Glazing	Emissivity	0.92
	Transmittance	0.9
	Material	Stainless steel S280
	Coating	Dark black surface
Aboorbor	Thickness (m)	0.001
Absorber	Thermal Conductivity (W/(m.K))	50
	Absorptivity	0.9
	Emissivity	0.9
	Material	Copper
	Running Length (m)	57.5
	Length of each Segment (m)	2.5
Circulation System	Number of Segments	23
-	Outer diameter (m)	0.01
	Inner diameter (m)	0.008
	Thermal Conductivity (W/(m.K))	300
	Material	Celotex
Back Insulation	Thickness (m)	0.1
	Thermal Conductivity (W/(m.K))	0.022
	Type of lamps	Quartz-halogen
Solar Simulator	Number of lamps	12
Solal Simulator	Radiant heat output per lamp (kW)	1
	Distance from collector (m)	2

8



205 Fig. 2 Physical model illustration of the shell-and-tube heat exchanger with multi-tubes and

206 longitudinal fins based LHS unit and vertical and radial positioning of thermocouples in shell container

207 [22].

## Table 2

Technical specifications of shell-and-tube heat exchanger based LHS unit used in test rig [21, 22]

Components	Parameters	Value
	Material	Copper
	Height (m)	0.385
Shell	Outer diameter (m)	0.450
	Inner diameter (m)	0.448
	Volume of shell (m <sup>3</sup> )	0.0607
	Material	Copper
	Height of each tube pass (m)	0.320
	Number of passes	21
Tube	Outer diameter (m)	0.022
	Inner diameter (m)	0.020
	Running length (m)	8.2
	Volume of tube (m <sup>3</sup> )	0.00312
	Material	Copper
	Height (m)	0.230
Fine	Length (m)	0.04
FIIIS	Thickness (m)	0.0015
	Number of fins	76
	Volume of fins (m <sup>3</sup> )	0.00105
Inculation	Material	CFC-free envirofoam
	Thickness (m)	0.05
	HTF Material	Water
	PCM Material	Paraffin (RT44HC)
PCM and HTF	Mass of PCM (kg)	40
	Packing factor of PCM	0.824 (Solid) 0.942 (Liquid)

## Table 3

Thermo-physical characteristics of paraffin as PCM, copper as construction material for shell-and-tube heat exchanger and water as HTF [42, 43]

Parameters	Paraffin (RT44HC)	Copper	Water
Phase change temperature (°C)	41-44	1083	0-4
Latent heat capacity (kJ/kg)	255	0.205	333.5
Density (kg/m <sup>3</sup> )	800 at 20 °C	8960 at 25 °C	999.93 at 5 °C
Density (kg/m²)	700 at 80 °C	8935 at 70 °C	977.75 at 70 °C
Specific heat capacity (k l/(kg K))	2.0	0 385	4.204 at 5 °C
Specific fleat capacity (KS/(Kg.K))	2.0	0.000	4.189 at 70 °C
Thermal conductivity $(M/(mK))$	0.2	308	0.571 at 5 °C
	0.2	390	0.663 at 70 °C
Coefficient of thermal expansion $(1/k)$	3 x 10-4	0 167 v 10-4	0.114 x 10⁻⁴ at 5 °C
Coefficient of thermal expansion (1/K)	3 X 10	0.107 × 10	5.71 x 10 <sup>-4</sup> at 70 °C

## 208 2.2. Experimental procedure

Prior to conducting charging cycles, the low temperature water is circulated through the serpentine tubing in the solar collector and multi-tube passes in the LHS unit to ensure good baseline initial temperature and to release trapped air in the loop. Initial temperature of the paraffin in the shell container of the LHS unit is maintained at 10 °C. Three-way valves are adjusted to regulate flow direction between the solar collector and the LHS unit for operating closed loop charging cycles.

Solar simulators are used to direct radiant heat at the solar collector. Water absorbs thermal 215 energy while passing through the serpentine tubing in the solar collector and therefore, the 216 temperature of water is increased from initial ambient to the selected range. The high 217 218 temperature water at the outlet of the solar collector is directed to pass through multi-tube 219 passes in the LHS unit where thermal energy is transferred to low temperature paraffin. Paraffin absorbs thermal energy and phase transition from solid to liquid starts. The low 220 temperature water at the outlet of the LHS unit is pumped back to the solar collector to 221 continue closed loop charging cycle. Charging cycle is continued until all fifteen 222 thermocouples in the shell container of the LHS unit register temperature higher than phase 223 change temperature of the paraffin. 224

Likewise, in the case of discharging cycles, it is ensured that the entire mass of paraffin is in liquid state and the thermocouples installed at top position at all zones in the shell container register paraffin temperature equal to 52 °C. Three-way valves are adjusted again to regular flow direction for open loop discharging cycle between the water supply and the LHS unit. The solar collector and centrifugal circulation pump are bypassed in discharging cycles.

230 In discharging cycles, the low temperature water is directed to pass through the LHS unit. Heat transfer occurs between low temperature water in multi-tube passes and high 231 temperature paraffin in the shell container. Paraffin enthalpy drops and hence phase 232 transition initiates from liquid to solid state. High temperature water at the outlet of the LHS 233 234 unit can be consumed and is desirable for domestic or industrial applications. The discharging cycle is continued until all fifteen thermocouples in the shell container register 235 temperature lower than the phase change temperature of paraffin and the temperature 236 237 gradient between the inlet and outlet of the water is less than 5 °C. The range of experimental tests conducted on charging and discharging cycles at varied operating 238 239 conditions of temperature and volume flow rates is presented in Table 4.

<b>U</b>	Sot of	Inlet	Statan	Reynolds No			
	Experiments	(°C)	No	1.5 l/min	2.0 l/min	2.5 l/min	3.0 l/min
	1-4	52	0.078	2950	4000	4950	5900
jing	5-8	57	0.118	3200	4300	5350	6400
narç cle	9-12	62	0.157	3450	4600	5750	6900
5 S	13-16	67	0.196	3700	4900	6150	7400
бĽ							
schargir cle	17-20	5	0.290	1050	1395	1745	2100
	21-24	10	0.251	1220	1625	2030	2435
<u>ن ت</u>	25-28	15	0.212	1400	1865	2330	2800

 Table 4

 Range of charging/discharging cycles conducted at varied operating conditions

## 240 3. Thermal performance evaluation

The flow regime of water in multi-tube passes in the LHS unit at varied operating conditions is determined using Reynolds number (*Re*) as follow [44]:

$$Re = \frac{4\dot{m}}{\pi D_{T,in}\mu_w} \tag{1}$$

where  $\dot{m}$ ,  $\mu_w$  and  $D_{T,in}$  represent mass flow rate of water (kg/s), dynamic viscosity of water at inlet temperature (Pa. s) and inner diameter of tube (m). In case of phase transition, the ratio between sensible and latent heat is evaluated through Stefan number (*Ste*), as follow [45]:

$$Ste = \frac{C_{p,pcm} \,\Delta T_{Ste}}{L} \tag{2}$$

where  $C_{p,pcm}$ , *L* and  $\Delta T_{Ste}$  represent the specific heat capacity of paraffin (J/(kg.K)), latent heat capacity of paraffin (J/kg) and temperature gradient between phase transition temperature of paraffin and inlet temperature of water (°C), respectively. Table 4 presents the nature of flow regime and Stefan number for each experimental test.

250 3.1. Thermal energy and mean power evaluation

Thermal energy charge  $Q_c$  (J) and discharge  $Q_d$  (J) by the paraffin in the shell container is equated from the energy balance equation for inlet and outlet enthalpy change of water, as follows:

$$Q_{c} = \sum_{t=0}^{t_{c}} \dot{m} \left( \frac{C_{p,in} + C_{p,out}}{2} \right) (T_{in} - T_{out}) \Delta t$$
(3)

$$Q_{d} = \sum_{t=0}^{t_{d}} \dot{m} \left( \frac{C_{p,in} + C_{p,out}}{2} \right) (T_{out} - T_{in}) \Delta t$$
(4)

where  $\dot{m}$ ,  $C_p$ ,  $T_{in}$ ,  $T_{out}$  and  $\Delta t$  represent the mass flow rate of water (kg/s), specific heat capacity of water at inlet and outlet (J/(kg.K)), temperature of water at inlet and outlet (°C) and time step to register data (s). Likewise, the instantaneous charging/discharging power can be equated by dividing Eq. 3 and Eq. 4 with  $\Delta t$ . Moreover, the average charging power  $\overline{P_c}$  (W) and discharging power  $\overline{P_d}$  (W) can be calculated as follows:

$$\overline{P}_c = \frac{Q_c}{t_c} \tag{5}$$

$$\overline{P_d} = \frac{Q_d}{t_d} \tag{6}$$

where  $t_c$  and  $t_d$  represent complete charging and discharging time of paraffin, respectively.

#### 260 3.2. Charging and discharging cycles efficiency

The charging and discharging cycle efficiencies indicators are introduced to evaluate thermal behaviour of the LHS unit at varied operating conditions, as follows:

$$\eta_c = \frac{Q_c}{Q_{c,max}} \tag{7}$$

$$\eta_d = \frac{Q_d}{Q_{d,max}} \tag{8}$$

where  $\eta_c$  and  $\eta_d$  symbolise charging and discharging efficiency, respectively.  $Q_{c,max}$  and  $Q_{d,max}$  represents maximum theoretical thermal energy charged/discharged by the LHS unit. As discussed earlier, the LHS unit is comprised of a shell-and-tube heat exchanger with longitudinal fins and paraffin as thermal storage material, therefore the overall thermal energy storage is equated as the collective thermal energy stored by heat exchanger and paraffin, as follows:

$$Q_{c,max} = Q_{HX} + Q_{PCM} \tag{9}$$

$$Q_{HX} = m_{HX}C_{p,HX}(T_i - T_f)$$
<sup>(10)</sup>

$$Q_{PCM} = \int_{T_i}^{T_m} m_{pcm} C_{p,pcm} dT + m_{pcm} L + \int_{T_m}^{T_f} m_{pcm} C_{p,pcm} dT$$
(11)

where  $m_{HX}$ ,  $C_{p,HX}$ ,  $T_i$  and  $T_f$  shows mass of copper used in heat exchanger (kg), specific heat capacity of copper (J/(kg.K)) and initial and final temperature of charging/discharging cycle (°C), respectively. Likewise,  $m_{pcm}$  and  $T_m$  represents mass of paraffin in the shell container (kg) and phase change temperature of paraffin (°C).

## 273 3.3. Effectiveness-NTU method

The design and thermal performance analysis are conducted by implementing effectiveness-NTU method. Effectiveness equates to the ratio between actual heat charged or heat discharged over maximum possible heat charged or heat discharged during phase change duration, as given [46]:

$$\varepsilon = \frac{T_{in} - T_{out}}{T_{in} - \overline{T_m}} \tag{12}$$

Eq. (12) evaluates instantaneous effectiveness over the period of phase change. In this article, the average transient temperature plot for paraffin is produced from all fifteen local temperatures and accordingly  $\overline{T_m}$  represents the average transient temperature of paraffin during phase change period. Due to transient nature of effectiveness, the output values are limited between 0 and 1. Similarly, the average charging/discharging effectiveness duringthe phase change period can be calculated as follows [38]:

$$\overline{\varepsilon_c} = \frac{1}{t_c} \sum_{t=0}^{t_c} \varepsilon_c \tag{13}$$

$$\overline{\varepsilon_d} = \frac{1}{t_d} \sum_{t=0}^{t_d} \varepsilon_d \tag{14}$$

Average effectiveness is correlated to average number of transfer units (NTU) of the LHS unit, which evaluates the average thermal resistance to heat transfer between water and phase change front of paraffin, as given [47]:

$$\bar{\varepsilon} = 1 - \exp(-NTU) \tag{15}$$

$$\bar{\varepsilon} = 1 - \exp\left(\frac{-UA_{ext}}{\dot{m}C_p}\right) \tag{16}$$

where *U* and  $A_{ext}$  indicate overall heat transfer coefficient (W/(m<sup>2</sup>.K)) and effective external heat transfer surface area (m<sup>2</sup>), respectively. Eq. (16) indicates average effectiveness as a function of mass flux. Further details regarding average effectiveness and mass flux relation could be found in [38, 39, 48, 49]. Similarly, a generalised correlation of Eq. (16) is established as follows:

$$\bar{\varepsilon} = 1 - \exp\left(-C.\left(\frac{A_{ext}}{\dot{m}}\right)\right) \tag{17}$$

where *C* is a constant that specifies heat charging/discharging performance of the LHS unit.

293 3.4. Evaluation of natural convection

To evaluate the impact of varying operating conditions on dominating mode of heat transfer between water and paraffin, the non-dimensional Rayleigh number (Ra) and Nusselt number (Nu) are implemented. Rayleigh number accounts for buoyancy driven flow, natural convection and nature of phase front moment (laminar or turbulent), as given [44]:

$$Ra = Gr_L Pr = \frac{g\beta(T_s - \overline{T_m})L_c^3}{v\alpha}$$
(18)

where  $g, \beta, T_s, L_c, v$  and  $\alpha$  symbolise gravitational acceleration (m/s<sup>2</sup>), coefficient of thermal expansion of paraffin (1/K), surface temperature of the tube ( $T_s = (T_{in} + T_{out})/2$ ), characteristic height of tube with longitudinal fins (m), kinematic viscosity of paraffin (m<sup>2</sup>/s) and thermal diffusivity of paraffin (m<sup>2</sup>/s), respectively. Tubes surfaces with longitudinal fins are treated as vertical plates for calculating Ra. Therefore, the characteristic height is considered equivalent to the height of each tube pass i.e. 0.32 m. Transient kinematic viscosity of paraffin is determined by implementing the following relation [50]:

$$v = \frac{\mu}{\rho} = \frac{1}{\rho} \left( 0.001 exp \left( -4.25 + \frac{1790}{\overline{T_m}} \right) \right)$$
(19)

The nature of phase front moment of paraffin in shell container is laminar for Rayleigh number in the range of  $(10^4 \le Ra \le 10^9)$  and turbulent for  $(Ra \ge 10^9)$  [44]. Moreover, Churchill and Chu [51] derived a correlation for local Nusselt number and Rayleigh number, as follows:

$$Nu = \left\{ 0.825 + \frac{0.387Ra^{1/6}}{(1 + (0.492/Pr)^{9/16})^{8/27}} \right\}^2$$
(20)

309 Similarly, the average values for Rayleigh number and Nusselt number of paraffin in a shell 310 container over the period of charging cycles are calculated by follow relations:

$$\overline{Ra} = \frac{1}{t_c} \sum_{t=0}^{t_c} Ra$$
(21)

$$\overline{Nu} = \frac{1}{t_c} \sum_{t=0}^{t_c} Nu$$
(22)

An empirical correlation between  $\overline{Ra}$  and  $\overline{Nu}$  is generated based on experimental results, in the following form:

313

$$\overline{Nu} = C\overline{Ra}^n \tag{23}$$

## 314 **4.** Results and Discussion

315 4.1. Repeatability and uncertainty analysis

To ensure repeatability of experimental results, three charging cycles are conducted at 316 constant operating conditions of water. Inlet temperature and volume flow rate are set 317 constant to 62 °C and 1.5 l/min, respectively. Transient temperature profiles of paraffin in the 318 319 shell container acquired from thermocouples are averaged and plotted against charging 320 time, as shown in Fig. 3 (A). Likewise, the transient variations in temperature plots of inlet 321 and outlet temperatures of water are illustrated in Fig. 3 (B). It can be noticed that the average temperature profiles of paraffin and inlet/outlet temperature plots of water for all 322 three experiments are almost identical. The experimental results validate good reliability and 323 repeatability of thermal performance of LHS system with average standard deviation values 324 of 0.49, 0.38 and 0.52 for average temperatures of paraffin, inlet and outlet temperatures of 325 326 water, respectively.



**Fig. 3** Repeatability demonstration of average temperature of paraffin (A) and inlet/outlet temperatures of water (B) while charging at constant inlet temperature of 62 °C and volume flow rate of 1.5 l/min.

The experimental errors associated with the uncertainties of measuring thermocouples and flow meter readings are determined by implementing Kline-McClintock's model [52, 53]. Hence, the propagation of uncertainties in measured and calculated thermal performance parameters of LHS system are evaluated by identifying the uncertainties of all independent variables, as follow [52, 54]:

$$R = f(x_1, x_2, x_3, \dots, x_n)$$
(24)

$$\delta R = \sqrt{\left(\frac{\partial R}{\partial x_1}\delta x_1\right)^2 + \left(\frac{\partial R}{\partial x_2}\delta x_2\right)^2 + \left(\frac{\partial R}{\partial x_3}\delta x_3\right)^2 + \dots + \left(\frac{\partial R}{\partial x_n}\delta x_n\right)^2}$$
(25)

where *R* represents the thermal performance parameter which is influenced by several independent variable i.e.  $x_1, x_2, x_3, ..., x_n$ . Similarly,  $\delta R$  and  $\delta x_1, \delta x_2, \delta x_3, ..., \delta x_n$  symbolise the uncertainty of thermal performance parameter and uncertainties of all independent variables, respectively. Thus, the maximum percentage uncertainties propagated in the experimental results of thermal energy charged and discharged are 6.1% and 4.8%, in effectiveness of charging and discharging cycles are 3.0% and 3.1%, in Rayleigh number and Nusselt number are 4.3% and 5.6%, respectively.

## 342 4.2. Charging power and energy stored

To conduct charging cycles, the high temperature water from the solar collector is directed to pass through multi-tube passes where thermal energy is transferred to low temperature solid paraffin in the shell container. Due to thermal energy storage, solid paraffin engages in phase transition to a liquid state. Transient local temperature responses of the paraffin are recorded by fifteen thermocouples installed in the shell container and are extensively discussed in [22]. Table 4 represents the set of sixteen charging cycles conducted at varied operating temperatures and volume flow rates.

Transient charging power of paraffin is evaluated by registering instantaneous rate of 350 change in enthalpy of water at inlet and outlet, as presented in Fig. 4. Transient charging 351 power can be categorised into three stages: rapid increase at initial stages until it reaches 352 the maximum value, brief sharp declination and gradual declination or almost uniform 353 charging power up until the end of charging cycle. It can be construed that conduction heat 354 transfer is the dominant mode of heat transfer at initial stages of the charging cycle. 355 Moreover, due to the relatively higher temperature gradient between the inlet water and the 356 paraffin at initial stages, the maximum value of charging power is reached. In the second 357 358 stage, the brief sharp declination demonstrates the onset of phase transition of the paraffin to a liquid state. In the third stage, natural convection starts dominating the charging 359 process. Furthermore, due to continued increase in paraffin temperature, the thermal 360 gradient between water and paraffin reduces which results in gradual decline in charging 361 362 power.

The impact of varying operating conditions on charging power is pronounced, as shown in 363 Fig. 4. It can be noticed from Fig. 4 (A) that at constant inlet temperature of 52 °C and 364 increasing volume flow rate from 1.5 l/min to 3 l/min, the maximum charging power is 365 augmented from 1.48 kW to 2.39 kW, respectively. Likewise, in the case of 57, 62 and 67 366 °C, the maximum charging power is improved from 1.38 – 1.98 kW, 1.54 – 1.97 kW and 2.15 367 - 2.63 kW, respectively (see Fig. 4 (B), (C) and (D)). Moreover, the average charging power 368 at varied operating conditions is illustrated in Fig. 4 (E). It is evident that with an increase in 369 operating inlet temperature of water, the average charging power is significantly improved. 370 For instance, with an increase in inlet temperature from 52 °C to 57, 62 and 67 °C and 371 372 constant volume flow rate of 3 l/min, the average power is enhanced by 22.39%, 75.53% and 76.41%, respectively. Furthermore, it is observed that the average charging power does 373 374 not exhibit significant enhancement with an increase in inlet temperature of the water from 375 62 °C to 67 °C.



Fig. 4 Transient and average charging power of LHS unit while conducting experimental chargingcycles at varied operating conditions.

Similarly, the accumulative thermal energy storage by paraffin in the LHS unit is assessed for all sixteen charging cycles at varied operating conditions, as presented in Fig. 5. It can be observed that the higher charging power at initial stages of charging cycle enables maximum accumulation of thermal energy. It is depicted by a linear increment in accumulative energy storage versus time. Further on, the charging power is continuously and gradually declining towards later stages of charging cycle which results in slower accumulation of thermal energy storage, as illustrated by curved increments.

386 Fig. 5 demonstrates the influence of varying operating conditions on accumulative thermal energy storage. It can be noticed that while charging at constant inlet temperature and 387 388 varying volume flow rate from 1.5 l/min to 3 l/min, the turbulent nature of water in the multitube passes increases which improves heat transfer to the paraffin in the shell container and 389 consequently, the rate of accumulative thermal energy storage is increased. Also, an 390 391 increase in volume flow rate of the water ensures a rise in average multi-tube wall 392 temperature. In the case of charging for 4 h at constant inlet temperature of 52 °C, the 393 accumulative energy storage is improved by 11.78%, 18.55% and 23.68% with an increase in volume flow rate from 1.5 I/min to 2, 2.5 and 3 I/min, respectively (see Fig. 5 (A)). 394 Similarly, while charging for 3 h at constant inlet temperature of 62 °C, the accumulative 395 energy storage is enhanced by 15.28%, 29.64% and 35.61%, respectively (see Fig. 5 (C)). 396 397 However, while charging for 2.5 h at constant inlet temperature of 67 °C, an augmentation of 398 a mere 3.03%, 6.11% and 10.61% is noticed, respectively (see Fig. 5 (D)). It is due to the fact that at higher inlet temperature, the average multi-tube wall temperature is insignificantly 399 influenced by increasing the volume flow rate of the water. Furthermore, the effect of 400 401 increasing inlet temperature and constant volume flow rate on thermal energy storage is 402 illustrated in Fig. 5 (E). It can be observed that while charging for 3 h, the amount of thermal energy charged is significantly improved from 7.42 MJ to 8.48, 11.07 and 12.76 MJ as the 403 inlet temperature is increased from 52 °C to 57, 62 and 67 °C, respectively. In other words, 404 an enhancement of 14.24%, 49.25% and 72.17% is achieved with an increase in inlet 405 406 temperature, respectively.

Therefore, an increase in inlet temperature of water has more pronounced impact on charging power and accumulative thermal energy storage of paraffin as compared to increasing volume flow rate. Hence, an increase in Stefan number has more significant impact on charging cycles as compared to increasing Reynolds number.



412 Fig. 5 Transient response of accumulative thermal energy storage during charging cycles at varied

413 operating conditions of inlet temperatures and volume flow rates.

## 414 4.3. Discharging power and energy released

During open loop discharging cycles, the low temperature water is channelled through multitube passes to retrieve thermal energy from high temperature liquid paraffin in the shell container. Due to heat transfer to water, the enthalpy of liquid paraffin reduces and thus phase transition to solid state occurs. Transient local temperature response of paraffin to discharging cycles are registered by the thermocouples in the shell container and are discussed in detail in [23]. The set of twelve discharging cycles at varied operating conditions are listed in Table 4.

Instantaneous discharging power of paraffin is computed from enthalpy gradient of water at 422 the inlet and outlet, as presented in Fig. 6. Similar to charging cycles, the time-wise 423 discharging power can be categorised into three stages: a rapid increase in discharging 424 power until maximum value is attained, a rapid but brief reduction in discharging power. 425 followed by almost constant gradual declination towards the end of the discharging cycle. At 426 427 initial stages of discharging cycle, the maximum discharging power is achieved due to higher temperature gradient between the water and paraffin. In the second stage, the sharp 428 429 reduction in discharging power indicates formation of the solidified layer of paraffin over multi-tube passes and longitudinal fins. Low thermal conductivity of solid paraffin combined 430 431 with restrained natural convection result in reduced discharging power. In the later stages, 432 an almost constant gradual declination in discharging power is noticed as a result of reduced 433 temperature gradient between the water and paraffin.

Moreover, the effect of varying operating conditions on discharging power is prominent. As 434 presented in Fig. 6 (A), while discharging at constant inlet temperature of 5 °C, the maximum 435 discharging power is augmented from 6.04 kW to 8.89 kW with an increase in volume flow 436 437 rate from 1.5 l/min to 3 l/min, respectively. Likewise, in the case of discharging at constant 438 inlet temperature of 10 °C and 15 °C, the maximum discharging power is improved from 5.75 439 - 9.56 kW and 5.18 - 10.37 kW with an increase in volume flow rate from 1.5 l/min to 3 I/min, respectively (see Fig. 6 (B) and (C)). Moreover, the influence of varying operating 440 conditions on average discharging power is illustrated in Fig. 6 (D). It can be noticed that 441 with an increase in temperature gradient by reducing inlet temperature from 15 °C to 10 and 442 5 °C and at constant volume flow rate of 3 l/min, the average discharging power is 443 significantly enhanced by a fraction of 14.18% and 27.04%, respectively. Similarly, in case of 444 constant inlet temperature of 10 °C, the average discharging power is augmented by 7.03%, 445 446 17.97% and 32.69% with an increase in volume flow rate from 1.5 l/min to 2, 2.5 and 3 l/min, respectively. 447

Further, the transient accumulative thermal energy retrieval from paraffin by water at varied 448 operating conditions are illustrated in Fig. 7. It can be noticed that due to higher discharging 449 power at the initial stages, the linear increment in thermal energy discharge is presented. 450 Afterwards, the accumulative thermal energy retrieval is slowed down as discharging power 451 452 declines which is indicated by curved increments. In the case of discharging at constant inlet 453 temperature of 5 °C for 1.5 h, the accumulative energy retrieved by water is significantly increased from 10.74 MJ to 13.64 MJ with an increase in volume flow rate from 1.5 l/min to 3 454 455 I/min, as shown in Fig. 7 (A). With an increase in volume flow rate, the thermal resistance offered to convective heat transfer of water in the tube passes is reduced and consequently 456 a higher amount of thermal energy is retrieved. Similarly, the rate of thermal energy 457 discharge is significantly enhanced by increasing temperature gradient between inlet water 458

and paraffin. As shown in Fig. 7 (D), the required time to discharge 12 MJ of thermal energy
is significantly reduced from 2.12 h to 1.82 h and 1.54 h as the inlet temperature is
decreased from 15 °C to 10 and 5 °C, respectively. Therefore, by adjusting the inlet
operating conditions, the desired temperature and power needs can be attained at the outlet
for utilisation in practical applications.





465 Fig. 6 Time-wise variation and average discharging power of the LHS unit while conducting466 discharging cycles at varied operating conditions.



468 **Fig. 7** Accumulative thermal energy retrieved from paraffin to water during discharging cycles at 469 varied operating conditions.

## 470 4.4. Energy efficiencies of the LHS system

Thermal efficiencies of the LHS unit for various charging and discharging cycles are given in 471 Table 5. During charging cycles, the range of charging efficiency increases with a rise in inlet 472 473 temperature from 52 °C to 62 °C. However, the enhancement is minimal for further increase in inlet temperature from 62 °C to 67 °C. The range of charging efficiency is caused by 474 475 imperfect thermal insulation of the LHS unit. Due to which some portion of thermal energy 476 captured by paraffin is lost to the surrounding air. Therefore, the paraffin keeps on receiving thermal energy from incoming hot water until a steady state is reached. Steady state differs 477 for various operating temperatures and thus the charging efficiency fluctuates [38]. In the 478 case of discharging cycles, a relatively minimal variation in the range of discharging 479 efficiency is noticed, with an average value of 84.92%. Furthermore, an increase in volume 480 481 flow rate has presented an insignificant impact on charging/discharging efficiencies [29, 55].

Table	5
	-

Thermal efficiencies of LHS unit during various charging/dischar	ging	cycles
------------------------------------------------------------------	------	--------

11101								
	Operating conditions	η <sub>c</sub> (%)		Operating conditions	η <sub>d</sub> (%)			
	$T_{in} = 52 \text{ °C}, \dot{V} = 2.5 l/min$	79.29	<i>(</i> )	$T_{in} = 05 \text{ °C}, \dot{V} = 2.0 \ l/min$	83.92			
	$T_{in} = 52 \text{ °C}, \dot{V} = 3.0 \ l/min$	81.97	les	$T_{in} = 05 \text{ °C}, \dot{V} = 3.0 \ l/min$	86.19			
les	$T_{in} = 62 \text{ °C}, \dot{V} = 1.5 l/min$	79.13	Š	$T_{in} = 10 ^{\circ}\text{C}, \dot{V} = 1.5  l/min$	85.68			
Š	$T_{in} = 62 \text{ °C}, \dot{V} = 2.0 \ l/min$	88.49	δ	$T_{in} = 10 ^{\circ}\text{C}, \dot{V} = 2.0  l/min$	87.05			
δ	$T_{in} = 62 \text{ °C}, \dot{V} = 2.5 l/min$	97.31	gir	$T_{in} = 10 ^{\circ}\text{C}, \dot{V} = 2.5  l/min$	87.58			
īgir	$T_{in} = 62 \text{ °C}, \dot{V} = 3.0 \ l/min$	99.10	hai	$T_{in} = 10 ^{\circ}\text{C}, \dot{V} = 3.0  l/min$	88.38			
haı	$T_{in} = 67 \text{ °C}, \dot{V} = 2.5 l/min$	98.69	isc	$T_{in} = 15 \text{ °C}, \dot{V} = 2.0 \ l/min$	80.82			
S	$T_{in} = 67 \text{ °C}, \dot{V} = 3.0 \ l/min$	99.75	Δ	$T_{in} = 15 \text{ °C}, \dot{V} = 3.0 l/min$	79.80			

#### 482 4.5. Effectiveness-NTU assessment of the LHS system

483 Average effectiveness provides better assessment of the specified design configuration of the LHS unit as compared to published data in the literature. It is measured during the phase 484 transition duration of charging/discharging cycles, whereas sensible period of heat 485 486 charge/discharge is neglected [48]. The sensitivity of average effectiveness to variable operating conditions of water is investigated during charging and discharging cycles, as 487 shown in Fig. (8) and Fig. (9). It can be noticed that with an increase in volume flow rate, the 488 489 average effectiveness for both charging and discharging cycles reduces. The maximum 490 reduction in average effectiveness during charging and discharging cycles are recorded as 40.24% and 31.81%, respectively. With an increase in volume flow rate, the heat transfer 491 coefficient is not proportionally increased [49]. Also, due to rapid circulation of water in multi-492 493 tube passes, less time is provided for proper heat transfer between the water and paraffin. 494 As a result, the gradient between inlet and outlet temperature is reduced and thus, the 495 average effectiveness is decreased. Furthermore, the impact of increasing inlet temperature 496 on average effectiveness during charging/discharging cycles is prominent. The maximum enhancement in average effectiveness is noticed to be 65.72% and 32.62% for charging and 497 498 discharging cycles, respectively.

Based on experimental calculation of average effectiveness at varied operating conditions, a non-linear regression technique is implemented to establish correlation between average effectiveness and mass flux at corresponding operating inlet temperature. The process flow diagram to develop correlations using non-linear regression technique is illustrated in Fig. 10. In the case of charging cycles, the correlation between average effectiveness and mass flux at respective inlet temperatures is given as follows:

$$\bar{\varepsilon}_{T_{in}=52^{\circ}C} = 1 - \exp\left(-0.01351 * \frac{A_{ext}}{\dot{m}}\right)$$
  $R^2 = 0.9574$  (26)

$$\bar{\varepsilon}_{T_{in}=57^{\circ}C} = 1 - \exp\left(-0.01679 * \frac{A_{ext}}{\dot{m}}\right)$$
  $R^2 = 0.9517$  (27)

$$\bar{\varepsilon}_{T_{in}=62^{\circ}C} = 1 - \exp\left(-0.02048 * \frac{A_{ext}}{\dot{m}}\right) \qquad R^2 = 0.9525$$
(28)

$$\bar{\varepsilon}_{T_{in}=67^{\circ}\text{C}} = 1 - \exp\left(-0.02439 * \frac{A_{ext}}{\dot{m}}\right) \qquad R^2 = 0.9239$$
(29)

505 Similarly, in case of discharging cycles, the correlations at respective three varied inlet 506 temperatures are developed as follows:

$$\bar{\varepsilon}_{T_{in}=05^{\circ}C} = 1 - \exp\left(-0.01181 * \frac{A_{ext}}{\dot{m}}\right)$$
  $R^2 = 0.9289$  (30)

$$\bar{\varepsilon}_{T_{in}=10^{\circ}\text{C}} = 1 - \exp\left(-0.01305 * \frac{A_{ext}}{\dot{m}}\right)$$
  $R^2 = 0.9403$  (31)

$$\bar{\varepsilon}_{T_{in}=15^{\circ}C} = 1 - \exp\left(-0.01699 * \frac{A_{ext}}{\dot{m}}\right)$$
  $R^2 = 0.9377$  (32)

507 It can be noticed that the relevant charging/discharging correlations resemble the 508 generalised correlation Eq. (17). Also, the constant C value is observed to be changing with 509 respect to inlet temperatures for both charging and discharging cycles. Therefore, governing 510 equations are developed for charging and discharging cycles to incorporate inlet 511 temperature of water  $T_{in}$  and phase change temperature of paraffin  $T_m$ , as follows:

$$\bar{\varepsilon}_c = 1 - \exp\left\{-(0.00072(T_{in} - T_m) + 0.0061) * \frac{A_{ext}}{\dot{m}}\right\} \qquad R^2 = 0.9546$$
(33)

$$\bar{\varepsilon}_d = 1 - \exp\left\{-(0.00052(\mathbf{T}_{in} - \mathbf{T}_m) + 0.0316) * \frac{A_{ext}}{\dot{m}}\right\} \qquad R^2 = 0.9281 \tag{34}$$

512 The governing equations for  $\bar{\varepsilon}_c$  and  $\bar{\varepsilon}_d$  enables prediction of the thermal performance of the

design configuration of the LHS unit for PCMs other than paraffin RT44HC. For this reason,
it is considered as an important development to determine the governing equation in
evaluating average effectiveness for a particular PCM in the desired temperature range.

evaluating average effectiveness for a particular POW in the desired temperature rai



516 517

Fig. 8 Average effectiveness of the LHS unit at various charging cycles





Fig. 9 Average effectiveness of the LHS unit at various discharging cycles



521 **Fig. 10** Process diagram to produce governing equation for effectiveness of the LHS unit during 522 charging and discharging cycles.

Further on, the thermal performance of the understudy design configuration is evaluated by 523 comparing charging effectiveness with designs published in the literature, as shown in Fig. 524 11. The corresponding designs from literature includes: horizontal single shell-and-tube heat 525 exchanger with paraffin by Hosseini et al. [56], four tubes-in-tank with salt hydrates by Tay et 526 al. [48], U-shaped tubes-in-tank with copper foam and paraffin composite by Meng and 527 Zhang [29], compact finned-tube heat exchanger with paraffin by Amagour et al.[38], 528 529 horizontal multi-tubes with transversal squared fins in rectangular container with organic 530 PCM by Gil et al. [57] and two copper pipes with four longitudinal fins in shell container with organic PCM by Murray and Groulx [45]. Fig. 11 illustrates that the current shell-and-tube 531 heat exchanger with multi-tube passes and longitudinal fins design configuration has better 532 thermal performance as compared to [38, 45, 48, 56, 57]. Therefore, it can be deduced that 533 design optimisation and better orientation of extended surfaces are essential to achieve 534 535 higher thermal performance. Moreover, the maximum charging effectiveness achieved by Meng and Zhang [29] is due to higher effective thermal conductivity of composite paraffin. 536





539 4.6. Natural convection characterisation

540 Thermal performance analyses of the longitudinal fins based LHS unit are conducted in 541 terms of dominant mode of heat transfer. Impact of natural convection on phase transition 542 process is evaluated by calculating Rayleigh number and Nusselt number from Eq. (18) -543 Eq. (23).

544 Transient response of Nusselt number to various charging cycles are illustrated in Fig. 12. At earlier stages, the Nusselt number peaks to maximum value due to higher charging power 545 546 and is then followed by a sharp decline. These earlier stages are conduction dominant and the decline is due to formation of liquefied layers of paraffin around tube passes and 547 longitudinal fins. Hence, the conduction heat transfer occurs through liquefied layers without 548 any provision for natural convection. However, as the charging cycle progress, the amount of 549 liquefied layers enlarge and buoyance driven natural convection intensifies. Due to 550 escalation in natural convection, a moderate increase in Nusselt number is noticed until a 551 552 local maxima is reached. As a result, the liquefied layers of paraffin ascend to upper region of shell container. In next stages, the accumulated liquefied layers of paraffin in upper region 553 recirculate to reach the solid paraffin in lower region. However, due to congestion and 554 stratification of liquefied paraffin in upper region, the influence of natural convection 555 556 weakens. Hence, the Nusselt number gradually decline until the end of charging cycle.

557 Fig. 13 illustrates the experimental calculated values of Nu-Ra for charging cycles at varied 558 operating conditions. Rayleigh numbers for charging cycles have ranged from  $5.63 \times 10^{10}$  – 1.37x10<sup>11</sup>, which indicates that the melt front movement in upward direction is turbulent in 559 nature. Moreover, the vertical orientation of longitudinal fins reinforces and supports natural 560 convection in paraffin. For higher Rayleigh number, the buoyant forces surpass viscous 561 forces and as a result, the Nusselt number and heat transfer coefficient are increased [58-562 60]. As a consequence, a higher charging rate with higher degree of thermal stratification of 563 564 paraffin in the shell container can be obtained.

565 Rayleigh number is significantly enhanced with an increase in inlet temperature. For instance, while charging at constant volume flow rate of 1.5 l/min, the Rayleigh number is 566 augmented by 32.25% with an increase in inlet temperature from 52 °C to 62 °C. Likewise, 567 the Nusselt number is improved from 678.99 to 751.56, which indicates that natural 568 convection is dominant mode of heat transfer at higher inlet temperatures. Conversely, it is 569 observed that an increase in volume flow rate reduces the temperature gradient between the 570 tube walls and the paraffin. Hence, the improvement in conduction heat transfer contributes 571 572 to reduction in both Rayleigh and Nusselt numbers. For instance, with an increase in volume 573 flow rate from 1.5 l/min to 3 l/min at constant inlet temperature of 62 °C, the Rayleigh and 574 Nusselt numbers are reduced by 23.57% and 9.23%, respectively.

575 Based on experimental data, an empirical correlation between Nu-Ra is developed using 576 non-linear regression technique, as follow:

$$\overline{Nu} = 0.122 * \overline{Ra}^{0.3404} \qquad 5.63 * 10^{10} \le \overline{Ra} \le 1.37 * 10^{11}$$
(35)

577 The coefficient of determination  $R^2$  is 0.98, which illustrates excellent statistical fitting of 578 regression line. In Eq. (35), the regression equation constant C and exponent *n* are in good 579 congruence with the results established for vertical plates by Bergman et al. [44] (C = 0.1 580 and n = 0.34).



## 581

582 **Fig. 12** Transient response of Nusselt number to charging cycles at constant inlet temperature and varying volume flow rates.



585 **Fig. 13** Nu-Ra correlation is developed from experimental charging cycles at varied operating 586 conditions.

587 4.7. Case studies of LHS system utilisation in practical applications

In order to implement the developed effectiveness correlations for design purposes in practical applications, two case studies are conducted. The effectiveness correlations can be implemented for a varied range of paraffin materials and operating conditions of water. However, in both case studies, the paraffin material and design features of the proposed LHS system are kept constant.

593 In the first case study, the viability of the proposed LHS systems in residential shower applications are assessed, as presented in Fig. 14 (A). As shown in Table 6, a number of 594 595 design assumptions are made such as: T<sub>in</sub> and T<sub>out</sub> are fixed to 10 °C and 35 °C which are average municipal water temperatures in winter and hot water demand for the shower, 596 respectively. The average flow rate of hot water during the shower is set to 5 l/min. Further 597 on, the average discharge power and effectiveness are evaluated by computing Eq. (4) and 598 Eq. (12), respectively. Subsequently, the design correlation Eq. (34) is implemented to 599 determine the required effective heat transfer area to sustain continuous supply of hot water 600 601 at the desired temperature. The calculated heat transfer area can be achieved by assembling four LHS systems in series to ensure hot water supply for service time of 60.97 602 min. The computed service time is sufficient for 6 people with an average shower time of 10 603 604 min.

In the second case study, the feasibility of the proposed LHS system integration with a 605 radiator is examined, as shown in Fig. 14 (B). It is assumed that the outlet temperature from 606 607 the LHS system is similar to the inlet temperature of radiators and vice versa. In order to sustain room temperature at 25 °C, the supply temperature to the radiator is set to 35 °C at a 608 flow rate of 2 l/min, as presented in Table 6. Based on design assumptions, the average 609 discharge power and effectiveness are calculated. It is deduced that a single LHS unit is 610 capable of achieving control room temperature for two radiators installed in parallel in two 611 separate rooms. Furthermore, the economic evaluations of the proposed LHS systems in 612 practical applications are conducted. Table 7 shows the capital cost of unit LHS system, 613 614 whereas, the payback periods of LHS systems as compared to electric and gas powered boilers are detailed in Table 8. To replace the electric boiler with the proposed LHS systems,

the payback periods for both case studies are 5.5 years and 5 years. Similarly, to replace the

gas powered boiler, the payback periods for both cases are 15.8 years and 8.1 years,

618 respectively.

## Table 6

Case studies of LHS unit implementation in domestic applications

Case Study	Т <sub>іп</sub> (°С)	T <sub>out</sub> (⁰C)	₿ (l/min)	Ż (kW)	ε <sub>exp</sub>	A <sub>ext,r</sub> (m²)	Required LHS units	t <sub>service</sub> (min)
6 x People Shower	10	35	5	8.69	0.735	7.94	4	60.97
2 x Parallel Radiators	25	35	2	1.39	0.526	0.94	1	68.51

619

 Table 7

 Capital cost of single LHS unit

Paraffin (RT44HC)	40 kg x 11.1 (£/kg)	£444.00
Copper Tank	100 L	£170.00
Copper Tube	8.2 m x 4.43 (£/m)	£36.33
Copper Fins	1.05 L x 151.74 (£/L)	£159.33
Welding		£15.99
Labour	8 h x 8.21 (£/h)	£65.68
Total capital cost (per unit)		£891.32



Fig. 14 Schematic representation of proposed LHS system utilisation in practical applications: (A) 1<sup>st</sup>
 case study and (B) 2<sup>nd</sup> case study.

Table 8

		Case Study # 01 Shower	Case Study # 02 Radiator
Energy required (kJ)		31856.83	5727.44
Energy including boiler efficiency (90%) (k	(J)	35042.51	6300.18
Total energy (kWh)		9.73	1.75
Scenario # 01			
Electric boiler cost (£/year)	15.75 (p/kWh)	£559.59	£100.61
Electricity standing charge (£/year)		£77.02	£77.02
Carbon dioxide factor (£/year)	0.305 (p/kWh)	£10.84	£1.95
Annual running cost (£/year)		£647.44	£179.57
Payback period (years)		5.5	5.0
Scenario # 02			
Gas boiler cost (£/year)	3.74 (p/kWh)	£132.88	£23.89
Gas standing charge (£/year)		£85.53	£85.53
Carbon dioxide factor (£/year)	0.184 (p/kWh)	£6.54	£1.18
Total annual running cost (£/year)		£224.95	£110.60
Payback period (years)		15.8	8.1

Payback period of proposed LHS system in domestic applications with two varied scenarios

# 622 5. Conclusion

623 In this article, experimental analysis is conducted on commercial grade paraffin (RT44HC) in a shell-and-tube heat exchanger with multi-tube passes and longitudinal fins based LHS 624 625 system. The LHS system is subjected to a series of sixteen close loop charging cycles with an integration to a flat plate solar collector and twelve open loop discharging cycles with a 626 627 connection to the municipal water supply at varied operating conditions. Thermal 628 performance of the designed LHS system is evaluated in terms of thermal energy 629 storage/retrieval, average charging/discharging power, charging/discharging thermal efficiencies, heat transfer characterisation and average effectiveness. Moreover, the 630 empirical correlations for charging/discharging effectiveness and Nu-Ra are derived to 631 632 enable evaluation of the practical utilisation of the designed LHS system in domestic and commercial applications. The following specific conclusions are drawn from this 633 634 experimental study:

635 During charging/discharging cycles, the transient power is categorised into three • stages as rapid increase until a maximum value is achieved, a brief rapid reduction is 636 followed by a uniform and gradual reduction until the end of charging/discharging 637 cycle. Likewise, the impact of increasing Stefan number on both charging/discharging 638 power and accumulative thermal energy storage/retrieval is more pronounced as 639 640 compared to increasing Reynolds number. For instance, the accumulative thermal energy storage is enhanced by 35.61% while charging at constant inlet temperature of 641 62 °C and increasing volume flow rate of 1.5 l/min to 3.0 l/min. Whereas, the 642 accumulative thermal energy is augmented by 49.25% while charging at constant 643 volume flow rate of 2 l/min and increasing inlet temperature from 52 °C to 62 °C. 644 Furthermore, the inlet operating conditions can be regulated to achieve the desired 645 charging/discharging power. 646

647 increase in volume flow rate has presented minimal influence An on charging/discharging thermal efficiencies. However, the average effectiveness is 648 significantly reduced with an increase in volume flow rate. For instance, the maximum 649 reduction in charging and discharging average effectiveness are 40.24% and 31.81%, 650 respectively. Moreover, the average effectiveness is significantly enhanced with an 651 increase in temperature gradient for both charging and discharging cycles. Likewise, 652 the average effectiveness of the current proposed LHS system is comparatively better 653 than other several design configurations published in the literature. Also, the empirical 654 correlations are developed to enable estimation of charging and discharging 655 effectiveness for particular paraffin material and inlet temperature. 656

- 657 The impact of design configurations of the shell-and-tube heat exchanger with vertical orientation of longitudinal fins on natural convection is evaluated by calculating Nu-Ra 658 at varied operating conditions. Rayleigh number has ranged from 5.63x10<sup>10</sup> -659 1.37x10<sup>11</sup>, which signifies the turbulent nature of melt front movement in upward 660 direction. Moreover, it is noticed that with an increase in temperature gradient, the 661 Rayleigh number is significantly enhanced and consequently, the Nusselt number and 662 heat transfer coefficient are augmented. On the contrary, in the case of increasing 663 volume flow rate, the Nu-Ra and heat transfer coefficient are noticeably reduced. 664 Furthermore, the empirical correlation of Nu-Ra for the proposed shell-and-tube heat 665 666 exchanger with longitudinal fins is in good agreement with established correlation for 667 vertical plate.
- It is construed from case studies that the proposed LHS system can fulfil application 668 based thermal energy demands by adjusting operating conditions or by assembling 669 several units in series or parallel combinations. For instance, the effectiveness 670 correlation is implemented to evaluate the feasibility of hot water supply for domestic 671 usage and the integration of radiators for controlling the environment. The case study 672 indicates that by assembling four LHS systems in series, the hot water supply demand 673 for six people to shower can be fulfilled. Also, the integration of two parallel radiators 674 with a single LHS system can sustain a controlled environment for more than an hour. 675

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