Abstract

Purpose

The purpose of this work is to study the antifriction, antiwear and tribolayer formation properties of the trihexyltetradecylphosphonium bis(2,4,4-trimethylpentyl) phosphinate ionic liquid (IL) as additive at 1 wt% in two base oils and their mixtures, comparing the results with those of a commercial oil.

Design/methodology/approach

The mixture of the base oils used in the formulation of the commercial oil SAE 0W20 plus the IL was tested under rolling/sliding and reciprocating conditions to determine the so-called Stribeck curve, the tribolayer formation and the antifriction and antiwear behaviors.

Findings

The use of this ionic liquid as additive in these oils does not change their viscosity; improves the antifriction and antiwear properties of the base oils, making equal or outperforming these properties of the SAE 0W20; and the thickness and formation rate of the tribolayer resulting from the IL-surface interaction is highly dependent on the type of base oil and influence on the friction and wear results.

Originality

The use of this ionic liquid allows to replace partial or totally commercial antifriction and antiwear additives.

Keywords: phosphonium-derived ionic liquid; additive; antifriction; antiwear; SAE 0W20

1. Introduction

The ionic liquids (ILs) are being used for over 40 years in different applications (Li Chum *et al.*, 1975; Wilkes *et al.*, 1982). These molten salts have properties that make them ideal for using in lubrication: non-flammability, high thermal stability, high conductivity, high polarity, ashless character and miscible in organic compounds (Bermúdez *et al.*, 2009; Blanco, *et al.*, 2011; Somers *et al.*, 2013; Viesca *et al.*, 2010; Ye *et al.*, 2001). The metal surface-IL interaction leads to a tribolayer formation on the surface contributing to better both antifriction and antiwear performances (Blanco, *et al.*, 2011; Otero, *et al.*2014; Viesca *et al.*, 2013).

All the above-mentioned good lubricating properties of the ILs increased the interest for testing them as neat lubricant or as additive. The first works studied ILs with fluorine-containing anions (Battez *et al.*, 2009; Chen *et al.*, 2009; Liu *et al.*, 2002; Mu *et al.*, 2005; Phillips and Zabinski, 2004), but these anions produced corrosive and toxic products in presence of water, making possible their use only in free-water conditions. Despite the excellent properties of ILs as lubricant, their industrial application as base stock is limited by their high cost compared to synthetic- and mineral-based oils (Somers *et al.*, 2013). However, numerous studies have been carried out about their use as additive in lubricants and it has been observed that small quantities of IL improve significantly tribological behavior of the base oil (Zhou *et al.*, 2009). The current commercial availability of phosphonium-derived ILs has promoted their testing in various applications (Fraser and MacFarlane, 2009) and shown their better performance as neat lubricant than

imidazolium-based ILs and conventional oils (Liu *et al.*, 2006; Minami *et al.*, 2008, 2010). The better miscibility of these ILs in non-polar oils has led to test them as lubricant additive (Barnhill *et al.*, 2014; González *et al.*, 2016; Hernández Battez *et al.*, 2017; Otero, *et al.*, 2014; Qu *et al.*, 2014; Totolin *et al.*, 2014; Zhou *et al.*, 2014).

Previous works reported the physical-chemical properties and tribological behavior of the phosphonium-derived IL $[P_{6,6,6,14}][(iC8)_2PO_2]$ as additive in both a mineral oil (González *et al.*, 2016) and a polyalphaolefín (González *et al.*, 2019). When the $[P_{6,6,6,14}][(iC8)_2PO_2]$ was used as additive in mineral and synthetic oils improved antifriction behavior under rolling/sliding motion compared to $[P_{6,6,6,14}][BEHP]$ counterpart and mixtures containing ZDDP (zinc dithiophosphates). Meanwhile, the $[P_{6,6,6,14}][(iC8)_2PO_2]$ behaved worse as additive than ZDDP in friction results under reciprocating motion, the former equalled or outperformed the antiwear behaviour of the latter when the IL concentration was 0.5 or 1 wt%.

This work is focused on the performance of the $[P_{6,6,6,14}][(iC8)_2PO_2]$ as additive to a mixture of two non-polar oils which are used as base stock in the formulation of the commercial oil SAE 0W20. This research is important for lubricant industry in order to determine the feasibility of replacing partial or totally the ZDDP used in the aforementioned commercial oil.

2. Methodology

2.1 Base oils and ionic liquid

The polyalphaolefin (PAO 4) and the hydrocracked mineral oil (YUBASE) used as base oils and the commercial low viscosity engine oil SAE 0W20 were kindly provided by Repsol S.A. The trihexyltetradecylphosphonium bis(2,4,4-trimethylpentyl) phosphinate ([$P_{6,6,6,14}$] [(iC8)₂PO₂]) was obtained from IOLITEC GmbH.

For preparing the mixtures, 1 wt.% of the IL was added to each oil, reaching a phosphorous concentration of 700 ppm (below the limit of 800 ppm in motor oils). Besides, a blend of 37 vol.% of YUBASE and 62 vol.% of PAO 4 was also additised with 1 wt.% of the IL to be compared with the SAE 0W20 (formulated with the same blend of YUBASE and PAO 4).

2.2 Viscosity measurements

A Stabinger Viscometer SVM3001 was used for measuring viscosity of all lubricant samples (neat oils and blends) at temperatures ranging from 20 to 100 °C. The viscosity index of the samples was calculated from the viscosity values.

2.3 Tribological tests

The friction behavior of all lubricant samples was studied through two type of tests in a ball-on-disc configuration with a tribometer (Mini-Traction Machine) which permit controlling ball and disc independently (rolling/sliding motion). The steel specimens used were a ball (AISI 52100, Ra $\approx 0.01 \mu m$, 19.05 mm-diameter) and a disc (AISI 52100, Ra $\approx 0.01 \mu m$, 46 mm-diameter). The contact between the specimens was lubricated with 10 mL of the lubricant samples and the cleaning procedure employed with the specimens is the same to that described in (González *et al.*, 2019). The first type of test (Stribeck curve determination) was performed at 50% of slide-to-roll ratio (SRR), speed range of 2000-10 mm/s, 0.95 GPa of maximum pressure, and temperature range of 40-100 °C. Electrical contact resistance (ECR) and friction coefficient were measured during tests. The second type of test was made using the technique employed by (Kapadia *et al.*, 2007) for measuring the tribolayer formation due to the lubricant-surface reaction. The tests were performed at 0.66 GPa of maximum pressure, 150 mm/s of speed, 50% of SRR, 100 °C of temperature and 60 min-duration.

Additionally, tribological tests under reciprocating motion were made in a UMT-3 (Bruker) tribometer. The test condition used in this case was: 60 min-duration, 60 N-load (equivalent to 1.91 GPa of maximum pressure), 15 Hz of frequency, 4 mm of stroke length, room temperature and relative humidity of 20-30%. The volume of lubricant sample used was 4 mL. As upper specimen was used a steel ball (AISI 52100, 63 HRC of hardness, 9.5 mm-diameter, Ra $\approx 0.01 \mu m$) and as lower specimen was used a steel disc (AISI 52100, machined from annealed rod, with 190–210 HV₃₀ of hardness and Ra < 0.02 μm). During the tests, friction coefficient was determined and confocal microscopy was used for measuring wear on the lower specimen surface after reciprocating tribological tests. At least three tests were conducted for each lubricant sample. The cleaning procedure of the specimens before tests was the same used in (González *et al.*, 2016).

2.4 Surface characterization

Scanning electron microscopy (SEM) and energy dispersive spectroscopy (EDS) were utilized after reciprocating tests in order to study the worn surface regarding wear mechanisms and chemical composition.

3. Results and discussion

3.1 Viscosity

Fig. 1 shows that the commercial oil (SAE 0W20) has the highest viscosity values at all temperatures. The rest of lubricant samples show similar viscosity values from 40 °C onwards. Due to the low concentration

used, the addition of the IL hardly modifies viscosity of the base oils. Taking into account that the commercial oil SAE 0W20 is formulated using a mixture of YUBASE and PAO 4 as base stock, the additive package used is responsible for its higher viscosity.

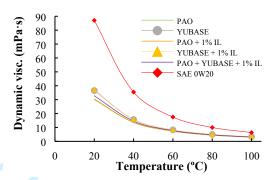


Fig. 1. Viscosity and density properties versus temperature.

3.2 Tribological tests

Fig. 2 shows the friction and ECR behavior against speed and temperature of the pure oils, their blends with the IL (separately and combined) and the commercial oil SAE 0W20 during the rolling/sliding tests. Friction decreased under elastohydrodynamic lubrication regime (high speed values) with temperature increase. These results are connected with the reduction of viscosity at increasing temperature and the consequent decrease in the lubricant film thickness. Due to film thickness is dependent on viscosity, the lower friction showed by the IL-containing mixtures at 40 and 60 °C regarding the SAE 0W20 is quite related to the above-mentioned viscosity results (Fig. 1). The change from elastohydrodynamic to mixed lubrication regime took place at higher speeds with temperature increase due to the greater viscosity decrease, which counteracts the influence of the speed on the lubricant film thickness reducing it and leads to more asperities contact between surfaces. Under mixed lubrication regime, the friction results are also controlled by the possible formation/destruction process of the tribolayer formed from the IL-surface interaction. Regarding the ECR results, it is also observed how temperature increase and/or lower entrainment speeds leads to thinner lubricant film thickness, a rise in asperities contact and hence lower ECR and higher friction values. In general, the highest ECR values were found in the elastohydrodynamic lubricant regime, where the lubricant film thickness (controlled by viscosity) avoid the asperities contact.

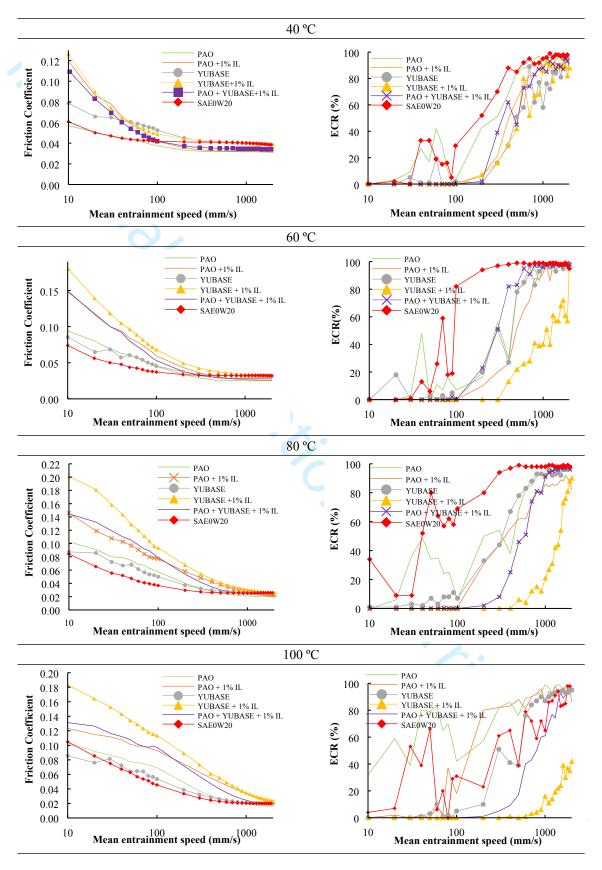


Fig. 2. Friction coefficient and ECR behaviors.

In addition, the measuring of tribolayer formation (Fig. 3) indicated a clear influence of the oil used. For the mixture of PAO 4 + 1 wt.% IL, it was noted a rapid tribolayer build up, with the thickest tribolayer measured after 60-min test. This result correlate with the high reactivity of this IL reported in previous works (González *et al.*, 2016, 2019). However, it seems that the YUBASE interferes in the tribolayer formation process when YUBASE + 1 wt.% IL was used, resulting in the thinnest tribolayer formed. For the mixture of YUBASE + PAO 4 + 1 wt.% IL, the tribolayer thickness formed was greater than that from YUBASE + 1 wt.% IL, but far away from the tribolayer formed when only the PAO 4 was used as base oil. Moreover, the SAE 0W20 shows a tribolayer formation rate almost constant, with a significant final thickness, although smaller than that from the PAO 4 + 1 wt.% IL. The interference images obtained during tribolayer formation tests can be shown in Fig. 4.

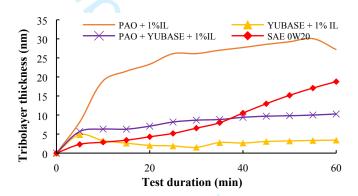


Fig. 3. Thickness of the tribolayer formed on the surface of the upper specimen (ball) during tests.

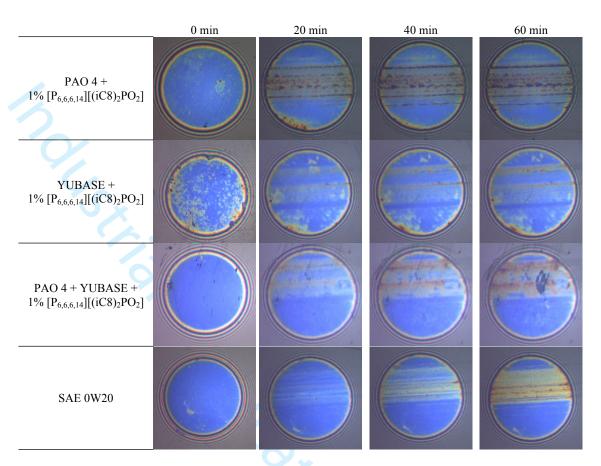


Fig. 4. Interference images of tribolayer formation on the surface of the ball during tests.

Fig. 5 shows the friction evolution during the 60-min test and the wear volume measured on the surface of the lower specimen (disc). With the exception of both pure PAO 4 and YUBASE, the remaining lubricant samples showed a relatively stable or decreasing friction coefficient evolution with testing time. The blend of PAO 4 +1 wt.% IL showed the lowest friction values, especially during the first half of the test, which can be connected with the rapid tribolayer formation showed by this mixture. In general, the additivation with the IL led to a decrease in friction compared to pure oils. In addition, a remarkable reduction in wear (comparable to that with SAE 0W20) was found with the additivation with the IL compared to the use of pure oils. This is also in accordance with the reactivity observed for this ionic liquid and its capacity of surface tribolayer formation.

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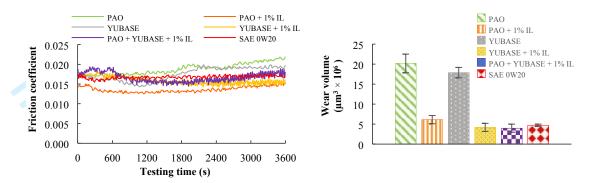


Fig. 5. Friction and wear from reciprocating tests.

3.3 Surface characterization

The SEM-EDS analysis shows how the wear track on the surfaces lubricated with pure oils are bigger than those in which the IL has been used as additive, Fig. 6. Plastic deformation was found at the edges of the wear track with a predominant adhesive wear mechanism in all cases. A difference morphology of the wear track can be noted when PAO 4 + 1 wt.% IL was used likely due to its higher capacity to form a tribolayer. In the EDS analysis, only the constituent elements of the AISI 52100 steel could be detected, with the A. where typ. exception of the surfaces lubricated with SAE 0W20 where typical elements included in the additive package (Zn, S, etc.) were also detected.

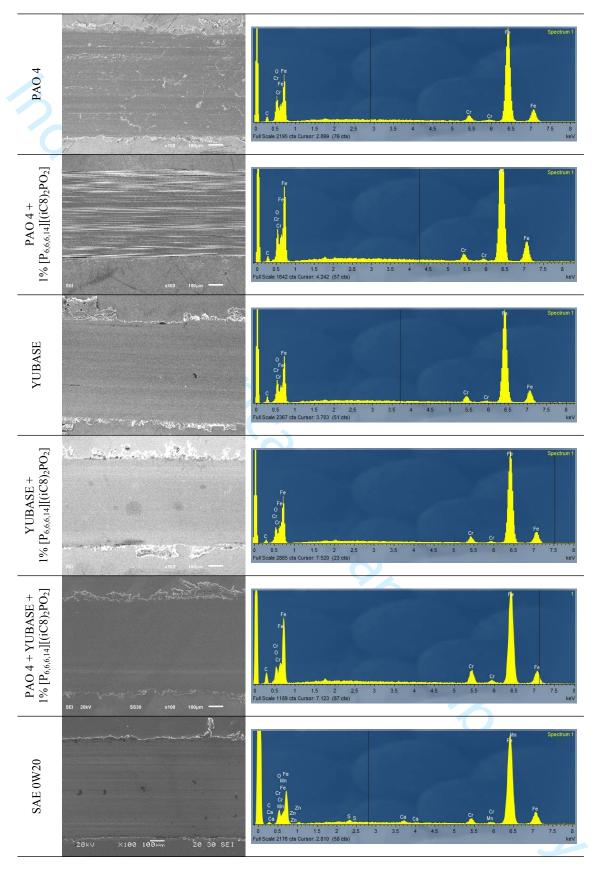


Fig. 6. SEM-EDS analysis after reciprocating tests.

4. Conclusions

The tribological performance of the trihexyltetradecylphosphonium bis(2,4,4-trimethylpentyl) phosphinate $([P_{6,6,6,14}][(iC8)_2PO_2])$ used as additive in two non-polar oils was compared with the commercial oil SAE 0W20. The main conclusions arose from this work are the following:

- The addition of [P_{6,6,6,14}][(iC8)₂PO₂] at 1 wt.% concentration practically does not change viscosity of the base oils.
- The use of [P_{6,6,6,14}][(iC8)₂PO₂] as an additive hardly changes friction under rolling/sliding motion but under reciprocating (sliding) motion reduces significantly both friction and wear.
- The formation rate of tribolayer from the IL-surface interaction is highly dependent on the base
 oil used and influence positively in friction or wear reductions.
- The use of [P_{6,6,6,14}][(iC8)₂PO₂] as additive improves the antifriction and antiwear behavior of the base oils, achieving antiwear results comparable to that of the commercial oil SAE 0W20.

References

- Barnhill, W.C. *et al.* (2014), "Phosphonium-organophosphate ionic liquids as lubricant additives: Effects of cation structure on physicochemical and tribological characteristics", *ACS Applied Materials and Interfaces*, Vol. 6 No. 24, pp. 22585–22593. https://doi.org/10.1021/am506702u.
- Battez, A.H. *et al.* (2009), "Tribological behaviour of two imidazolium ionic liquids as lubricant additives for steel/steel contacts", *Wear*, Vol. 266 No. 11–12, pp. 1224–1228, https://doi.org/10.1016/j.wear.2009.03.043.
- Bermúdez, M.-D. *et al.* (2009), "Ionic liquids as advanced lubricant fluids", *Molecules*, Vol. 14 No. 8, pp. 2888–2908, https://doi.org/10.3390/molecules14082888.
- Blanco, D. *et al.* (2011), "Lubrication of CrN coating with ethyl-dimethyl-2-methoxyethylammonium tris(pentafluoroethyl)trifluorophosphate ionic liquid as additive to PAO 6", *Tribology Letters*, Vol. 41 No. 1, pp. 295–302, https://doi.org/10.1007/s11249-010-9714-1.
- Blanco, D. *et al.* (2011), "Use of ethyl-dimethyl-2-methoxyethylammonium tris(pentafluoroethyl) trifluorophosphate as base oil additive in the lubrication of TiN PVD coating", *Tribology International*, Vol. 44 No. 5, pp. 645–650, https://doi.org/10.1016/j.triboint.2011.01.004.
- Chen, Y.M. et al. (2009), "The tribological performance of BCN films under ionic liquids lubrication", Diamond and Related Materials, Vol. 18 No. 1, pp. 20–26,

- https://doi.org/10.1016/j.diamond.2008.07.023.
- Fraser, K.J. and MacFarlane, D.R. (2009), "Phosphonium-based ionic liquids: An overview", *Australian Journal of Chemistry*, Vol. 62 No. 4, pp. 309–321, https://doi.org/10.1071/CH08558.
- González, R. *et al.* (2016), "Effectiveness of phosphonium cation-based ionic liquids as lubricant additive", *Tribology International*, Vol. 98, pp. 82–93, https://doi.org/10.1016/j.triboint.2016.02.016.
- González, R. *et al.* (2019), "Two phosphonium cation-based ionic liquids as lubricant additive to a polyalphaolefin base oil", *Journal of Molecular Liquids*, Vol. 293, p. 111536, https://doi.org/10.1016/j.molliq.2019.111536.
- Hernández Battez, A. *et al.* (2017), "Two phosphonium cation-based ionic liquids used as lubricant additive: Part I: Film thickness and friction characteristics", *Tribology International*, Vol. 107, pp. 233–239, https://doi.org/10.1016/j.triboint.2016.10.048.
- Kapadia, R. *et al.* (2007), "In situ observation of phosphorous and non-phosphorous antiwear films using a mini traction machine with spacer layer image mapping", *Tribology International*, Vol. 40 No. 10–12, pp. 1667–1679, https://doi.org/10.1016/j.triboint.2007.01.021.
- Li Chum, H. *et al.* (1975), "An Electrochemical Scrutiny of Organometallic Iron Complexes and Hexamethylbenzene in a Room Temperature Molten Salt", *Journal of the American Chemical Society*, Vol. 97 No. 11, pp. 3264–3265, https://doi.org/10.1021/ja00844a081.
- Liu, W. *et al.* (2002), "Tribological performance of room-temperature ionic liquids as lubricant", *Tribology Letters*, Vol. 13 No. 2, pp. 81–85, https://doi.org/10.1023/A:1020148514877.
- Liu, X. *et al.* (2006), "Tribological performance of phosphonium based ionic liquids for an aluminum-on-steel system and opinions on lubrication mechanism", *Wear*, Vol. 261 No. 10, pp. 1174–1179, https://doi.org/10.1016/j.wear.2006.03.018.
- Minami, I. *et al.* (2010), "Tribo-chemistry of phosphonium-derived ionic liquids", *Tribology Letters*, Vol. 40 No. 2, pp. 225–235, https://doi.org/10.1007/s11249-010-9626-0.
- Minami, I. *et al.* (2008), "The tribological properties of ionic liquids composed of trifluorotris(pentafluoroethyl) phosphate as a hydrophobic anion", *Tribology Letters*, Vol. 30 No. 3, pp. 215–223, https://doi.org/10.1007/s11249-008-9329-y.
- Mu, Z. et al. (2005), "Effect of the functional groups in ionic liquid molecules on the friction and wear behavior of aluminum alloy in lubricated aluminum-on-steel contact", *Tribology International*, Vol.

- 38 No. 8, pp. 725–731, https://doi.org/10.1016/j.triboint.2004.10.003.
- Otero, I. *et al.* (2014), "Friction and anti-wear properties of two tris(pentafluoroethyl) trifluorophosphate ionic liquids as neat lubricants", *Tribology International*, Vol. 70, pp. 104–111, https://doi.org/10.1016/j.triboint.2013.10.002.
- Otero, I. et al. (2014), "Ionic liquids based on phosphonium cations As neat lubricants or lubricant additives for a steel/steel contact", ACS Applied Materials and Interfaces, Vol. 6 No. 15, pp. 13115–13128, https://doi.org/10.1021/am502980m.
- Phillips, B.S. and Zabinski, J.S. (2004), "Ionic liquid lubrication effects on ceramics in a water environment", *Tribology Letters*, Vol. 17 No. 3, pp. 533–541, https://doi.org/10.1023/B:TRIL.0000044501.64351.68.
- Qu, J. et al. (2014), "Comparison of an oil-miscible ionic liquid and ZDDP as a lubricant anti-wear additive", *Tribology International*, Vol. 71, pp. 88–97, https://doi.org/10.1016/j.triboint.2013.11.010.
- Somers, A.E. *et al.* (2013), "A review of ionic liquid lubricants", *Lubricants*, Vol. 1 No. 1, pp. 3–21, https://doi.org/10.3390/lubricants1010003.
- Totolin, V. *et al.* (2014), "Lubrication mechanism of phosphonium phosphate ionic liquid additive in alkylborane-imidazole complexes", *Tribology Letters*, Vol. 53 No. 2, pp. 421–432, https://doi.org/10.1007/s11249-013-0281-0.
- Viesca, J.L. *et al.* (2010), "Assessing boundary film formation of lubricant additivised with 1-hexyl-3-methylimidazolium tetrafluoroborate using ECR as qualitative indicator", *Wear*, Vol. 269 No. 1–2, pp. 112–117, https://doi.org/10.1016/j.wear.2010.03.014.
- Viesca, J.L. *et al.* (2013), "FAP⁻ anion ionic liquids used in the lubrication of a steel-steel contact", *Tribology Letters*, Vol. 52 No. 3, pp. 431–437, https://doi.org/10.1007/s11249-013-0226-7.
- Wilkes, J.S. *et al.* (1982), "Dialkylimidazolium Chloroaluminate Melts: A New Class of Room-Temperature Ionic Liquids for Electrochemistry, Spectroscopy, and Synthesis", *Inorganic Chemistry*, Vol. 21 No. 3, pp. 1263–1264, https://doi.org/10.1021/ic00133a078.
- Ye, C. *et al.* (2001), "Room-temperature ionic liquids: A novel versatile lubricant", *Chemical Communications*, Vol. 21, pp. 2244–2245, https://doi.org/10.1039/b106935g.
- Zhou, F. *et al.* (2009), "Ionic liquid lubricants: Designed chemistry for engineering applications", *Chemical Society Reviews*, Vol. 38 No. 9, pp. 2590–2599, https://doi.org/10.1039/b817899m.

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13301-13311, https://doi.org/10.1021/la50323.